

# **Mass Spectrometry in Forensic Science**

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## **Overview**

- Introduction to forensic sciences
- Uses of mass spectrometry in forensic sciences
- Typical instrumentation in forensic sciences
- Applications of new instrumentation

## Introduction to Forensic Sciences

Forensic Sciences is defined as: the application of a broad spectrum of **sciences** to answer questions of interest to the legal system.



## Introduction to Forensic Sciences

Typical analytical sections within a forensic science laboratory:

**Drug Chemistry** – Analysis of pills, powders, liquids, plant materials, and other suspicious items for illegal drug content

**Toxicology** – Analysis of biological samples for alcohol, prescription medication, drugs of abuse, and other chemicals that are not naturally occurring in the body

**DNA** – Extraction and amplification of DNA from biological fluids for identification

**Firearms** – Bullet pattern recognition and analysis of gun powder

**Fire Debris** -- Identification of ignitable liquids used in arsons

## Standards for Accepting the Scientific Validity of a Procedure, Technique, and Principle

- Alabama
  - Frye standard: the court must decide if the questioned procedure, technique, and principles are “generally accepted” by a relevant community
  - Federal Rule 702: a witness qualified as an expert may testify in the form of an opinion
- Federal
  - Daubert:
    - Has it been tested?
    - Has it been published and peer reviewed?
    - Potential rate of error
    - Existence and maintenance of standards controlling the techniques operation
    - Accepted in the relevant scientific community

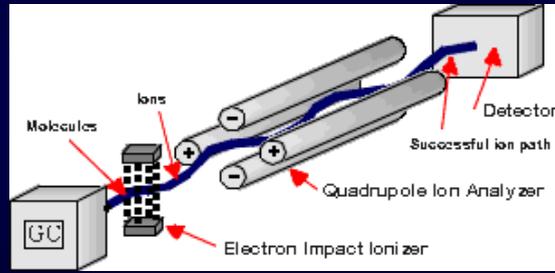
## Mass Spectrometry in Forensic Science

A gas chromatograph with a mass spec detector is the final tool used in the analysis of drug chemistry and toxicology samples for identification and confirmation.



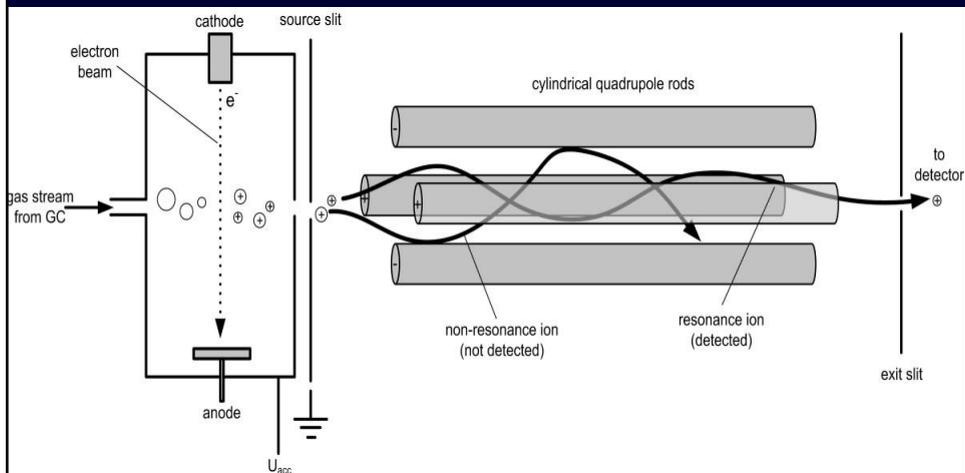
# Typical forms of Mass Spectrometry in Every Forensic Science Lab

## Gas Chromatography-Mass Spectrometry (GC-MS)



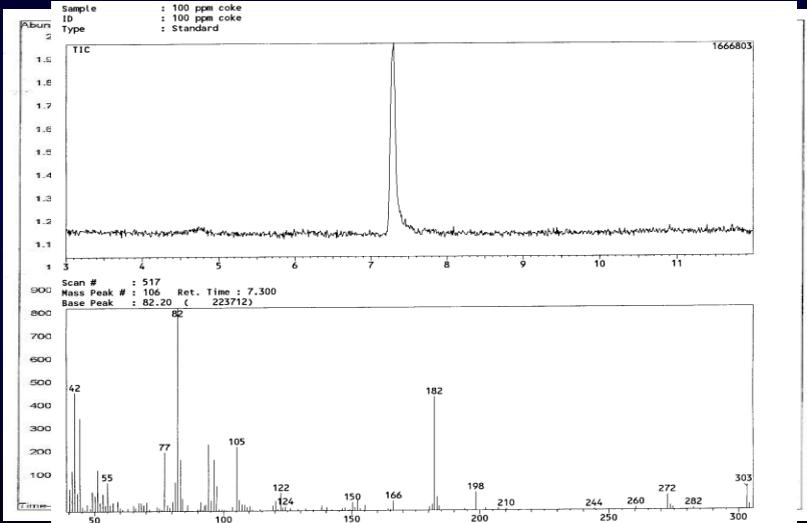
[http://www.chem.arizona.edu/massspec/intro\\_html/intro.html](http://www.chem.arizona.edu/massspec/intro_html/intro.html)

# Typical forms of Mass Spectrometry in Every Forensic Science Lab



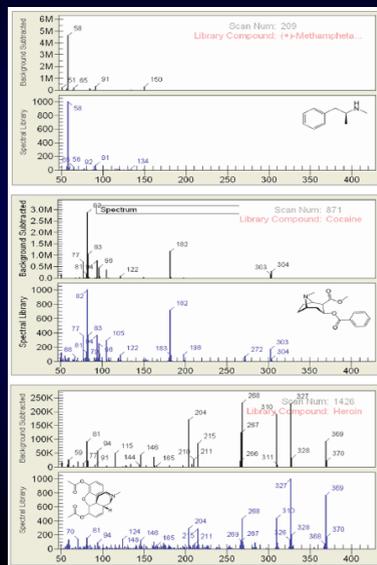
<http://www.microbialcellfactories.com/content/figures/1475-2859-6-6-4-1.jpg>

## Typical forms of Mass Spectrometry in Every Forensic Science Lab



## Typical forms of Mass Spectrometry in Every Forensic Science Lab

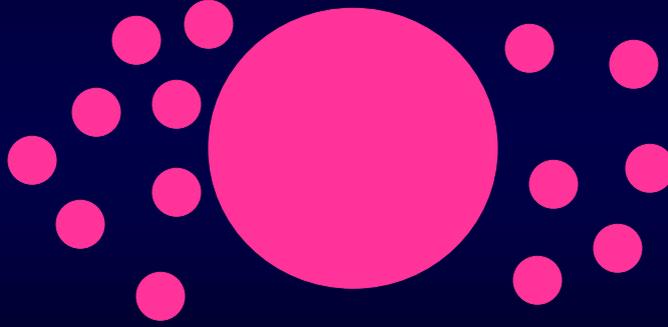
### Gas Chromatography-Mass Spectrometry (GC-MS) Spectrum



Spectra are searched against a library of known compounds in an effort to identify every peak in the TIC

A standard is analyzed on the instrument to generate a known retention time and spectrum of the compound for that instrument

## Problems Encountered with the GC/MS



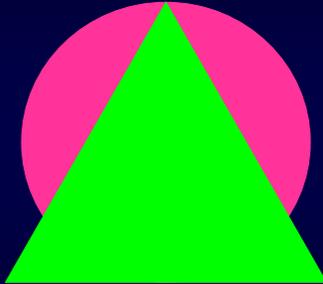
Lose the parent ion of the compound upon ionization  
in the instrument  
Example: Methadone

## Problems Encountered with the GC/MS



Derivatize the compound for analysis with GC/MS  
which decreases detection of low level compounds  
Example: THC

## Problems Encountered with the GC/MS



Heat labile compound will be identified as a related compound, but not the actual compound  
Example: Clorazepate to Nordiazepam

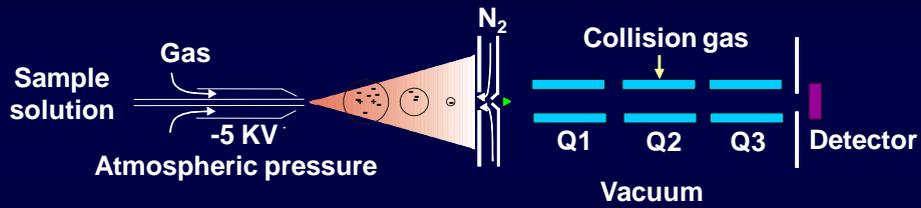
## New Technology

- Four new instruments have been brought into the department in October 2008
  - AccuTOF-DART mass spectrometer
  - 3200 QTRAP mass spectrometer with LC
  - 3200 QTRAP mass spectrometer with DART
  - HS-GC-MSD



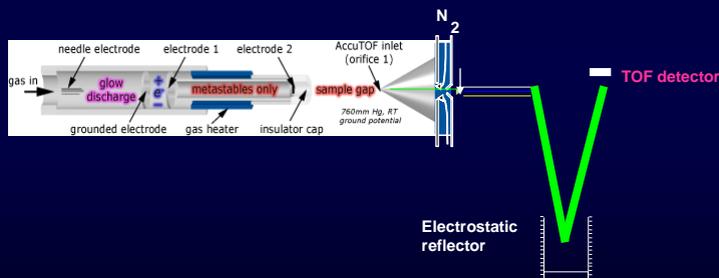
## Different forms of Mass Spectrometry

### Liquid Chromatography Electrospray Ionization Mass Spectrometry (LC-ESI-MS)



## Different forms of Mass Spectrometry

### Direct Analysis in Real Time with Time of Flight Mass Spectrometry

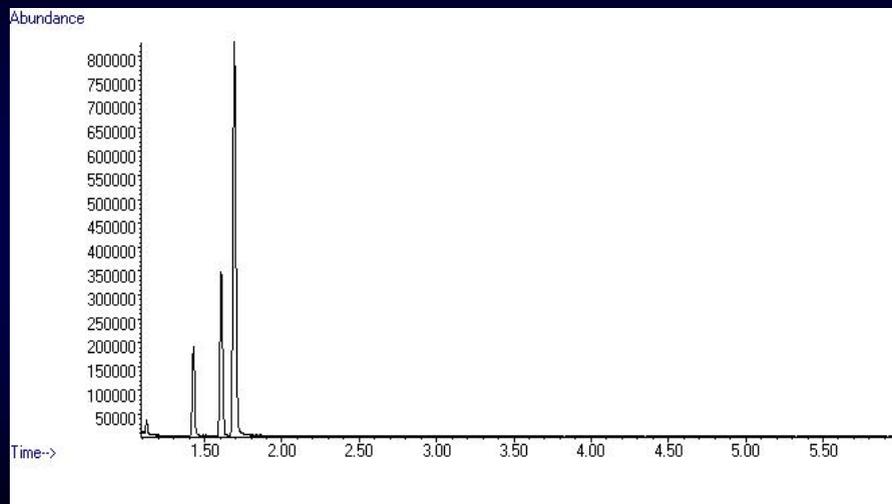


## HS-GC-MSD

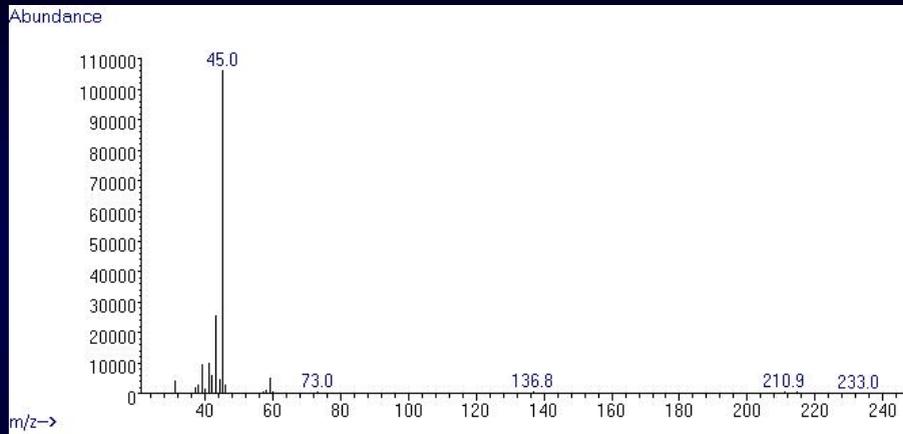
- This instrument provides opportunity for qualitative and quantitative identification of volatile compounds



## TIC of Volatiles Mix

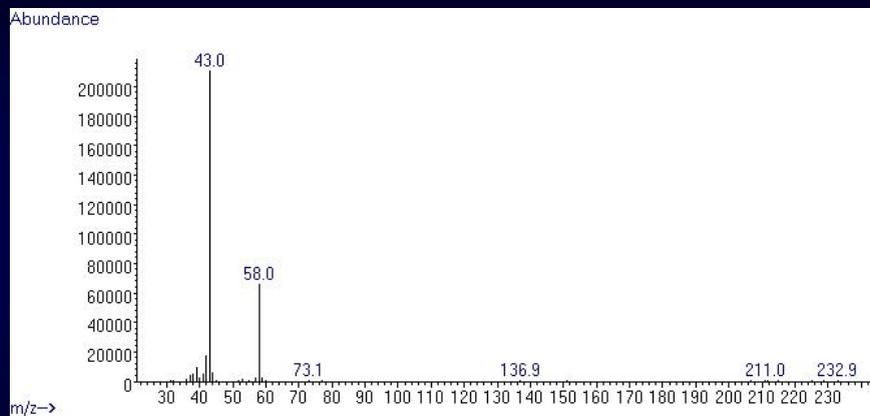


## Spectrum of Peak at 1.44 min



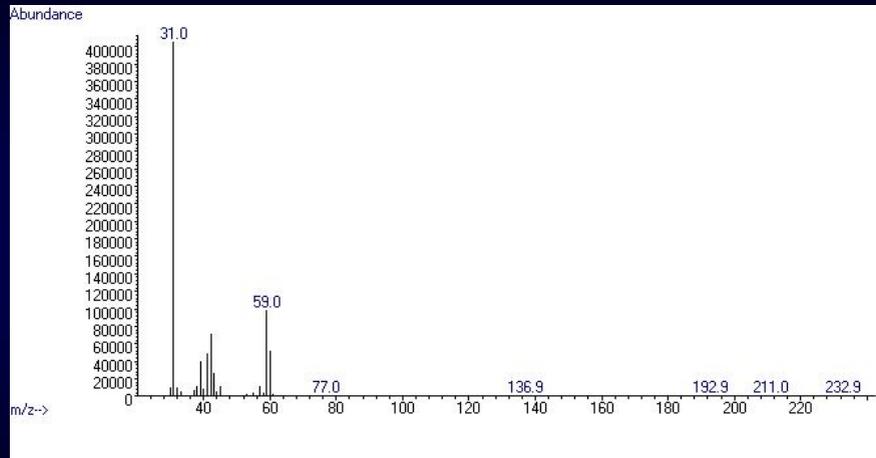
Isopropanol

## Spectrum of Peak at 1.61 min



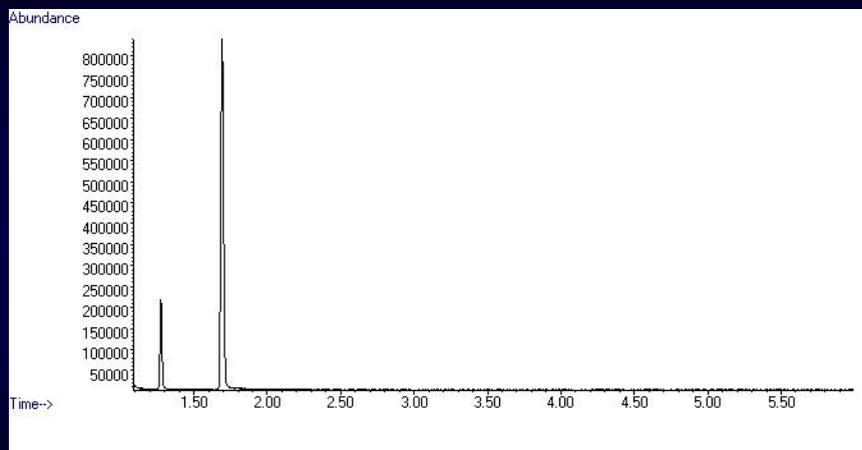
Acetone

## Spectrum of Peak at 1.70 min

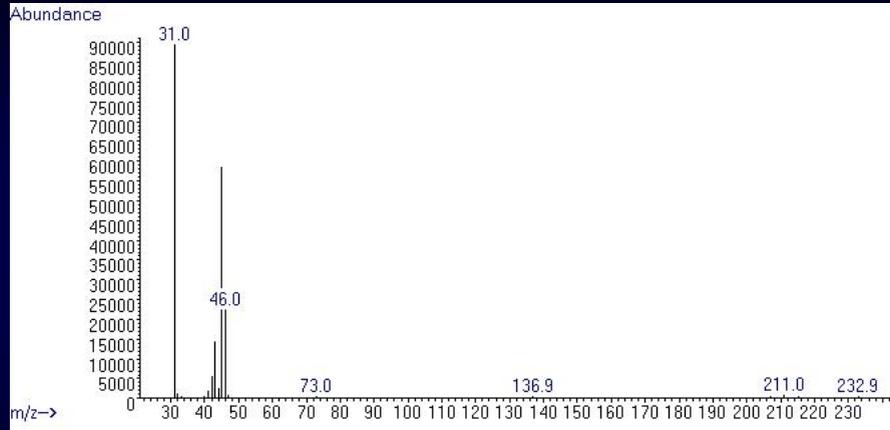


1-propanol (IS)

## TIC of Ethanol Standard

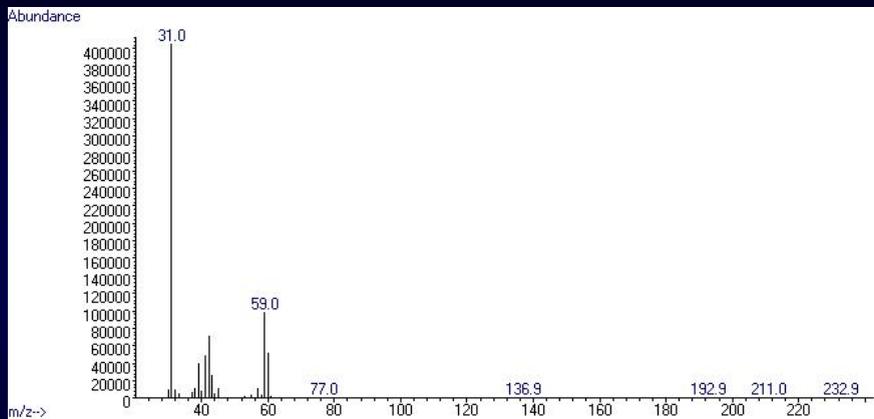


## Spectrum of Peak at 1.29 min



Ethanol

## Spectrum of Peak at 1.70 min



1-propanol (IS)

## Summary

- Method development is underway with the HS-GC-MSD
  - Good separation and spectra from the volatiles mix and ethanol standard
    - Ready to start validation
  - Developing method for commonly abused inhalants
  - Developing a screening for other volatile compounds
    - Example: GHB

## AccuTOF-DART MS

- The DART is the first open air, ambient ion source for a mass spectrometer
- Coupled to a time of flight instrument exact mass measurements can be used in the putative identification of compounds

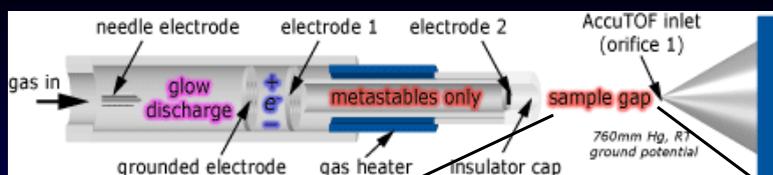


## 3200 QTRAP-DART MS

- Coupled to a hybrid triple quadrupole/Trap instrument molecular ions can be individually fragmented for identification of sample components



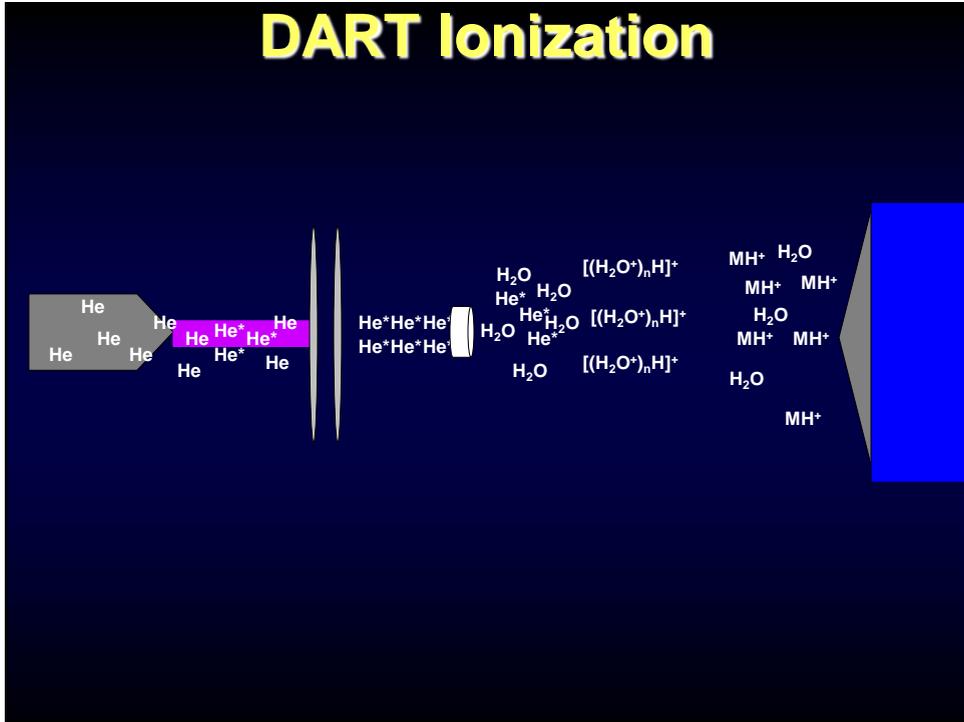
## DART Ionization



- Penning ionization: energy is transferred from metastable ions ( $M^*$ )
- Positive ions:  $He^*$  ionizes water which transfers a proton to the sample
- Negative ions: Penning electrons are rapidly thermalized and captured by oxygen which ionizes the sample

<http://www.jeolusa.com/PRODUCTS/AnalyticalInstruments/MassSpectrometers/AccuTOFDART/AccuTOFDARTIonizationMechanisms/tabid/450/Default.aspx>

# DART Ionization



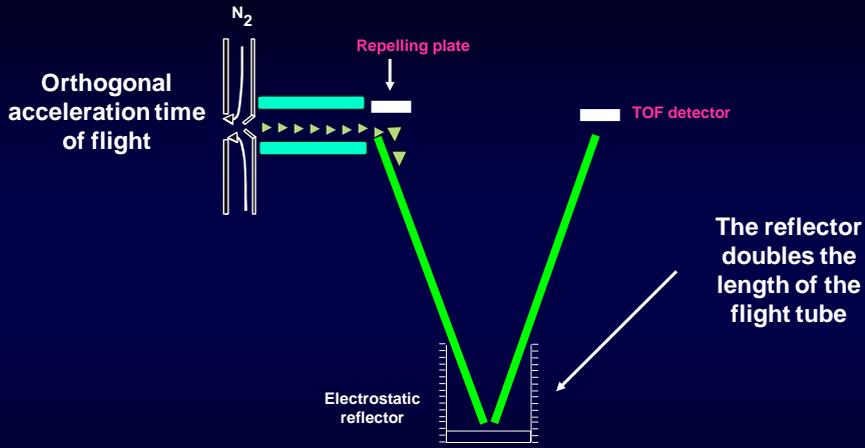
# Time of Flight Detector



$$t = (d/\sqrt{2U})((\sqrt{m/z}))$$

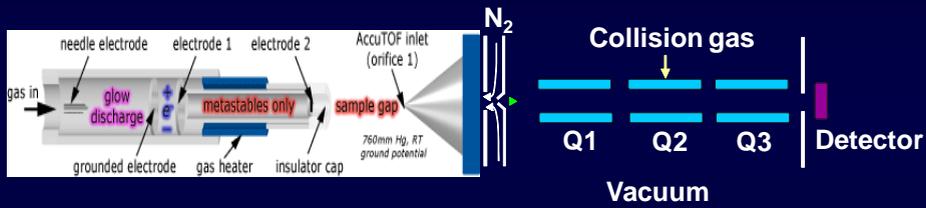
- t = time
- d = flight tube distance
- U = accelerating voltage
- m = mass
- z = charge

# AccuTOF Mass Spectrometer

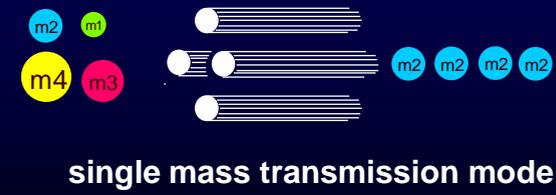


# Different forms of Mass Spectrometry

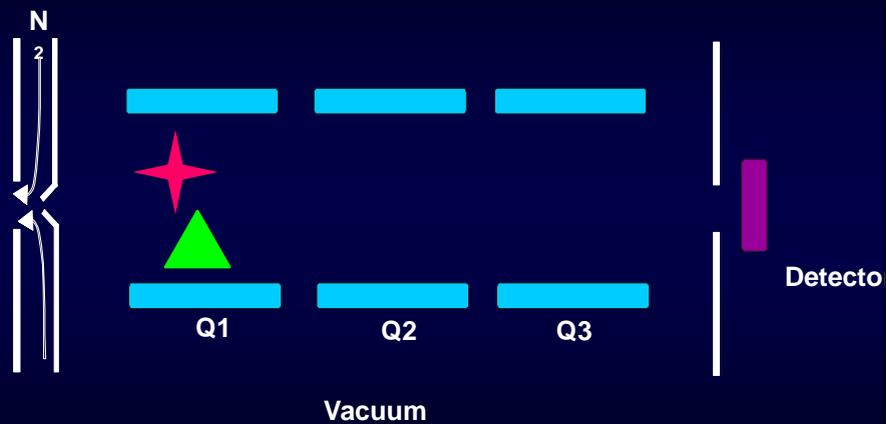
## DART Ionization Tandem Mass Spectrometry



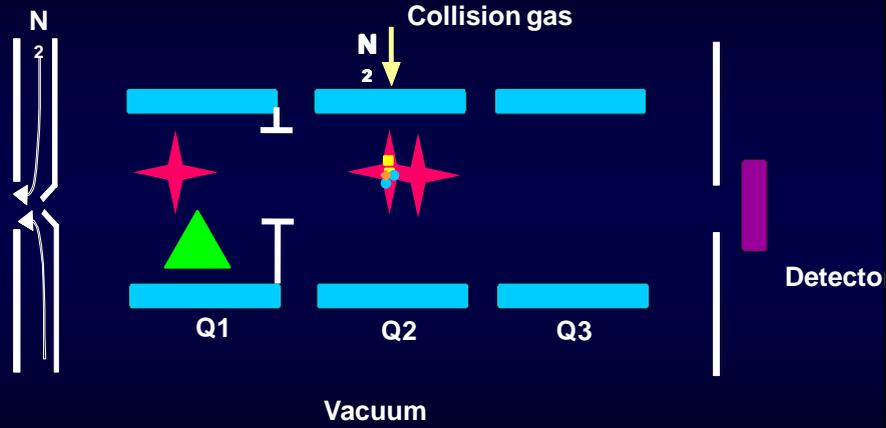
## Quadrupoles have variable ion transmission modes



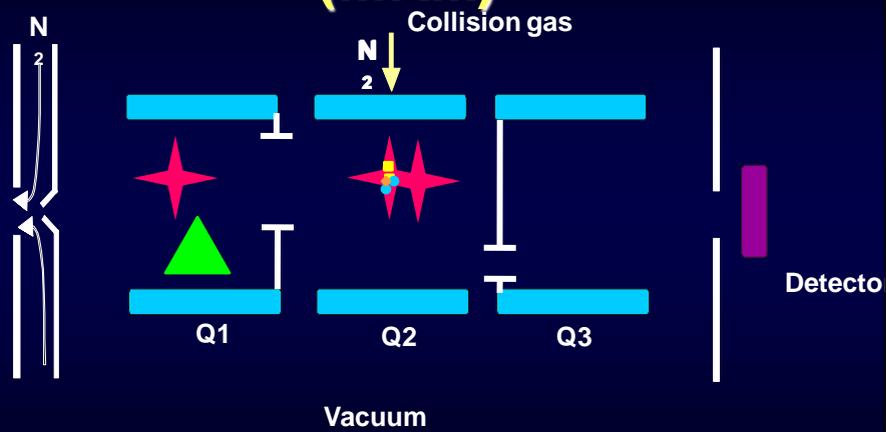
## Molecular Ion Scanning



## Product Ion Scanning



## Multiple Reaction Monitoring (MRM)



## Sample Introduction with the AccuTOF-DART MS



Liquid samples are introduced with a glass capillary tube closed at one end

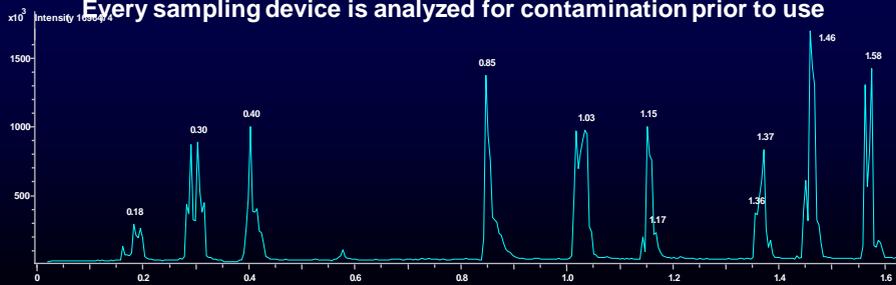


Solid samples are introduced into the stream with tweezers

## Sample Introduction with the AccuTOF-DART MS



Every sampling device is analyzed for contamination prior to use



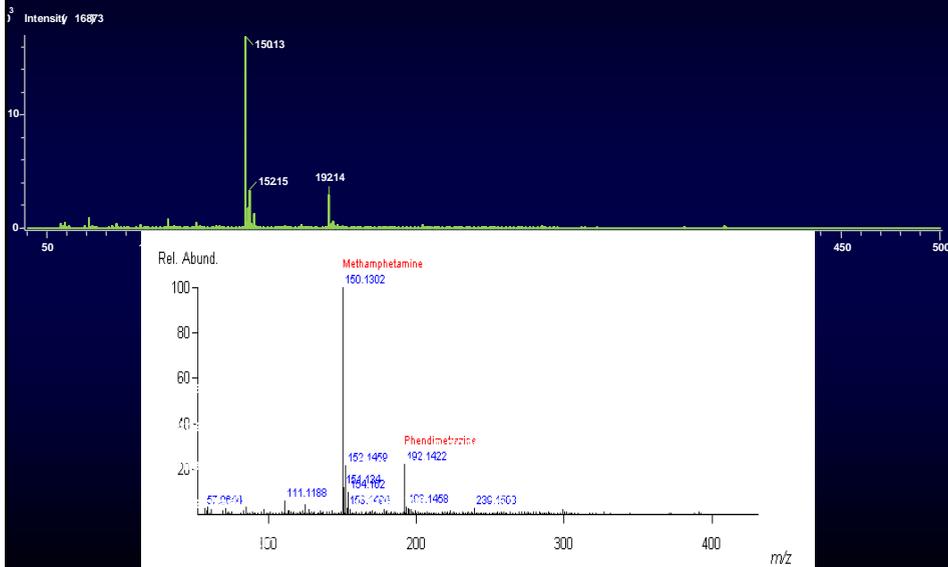
## Types of Samples Analyzed with the AccuTOF-DART MS

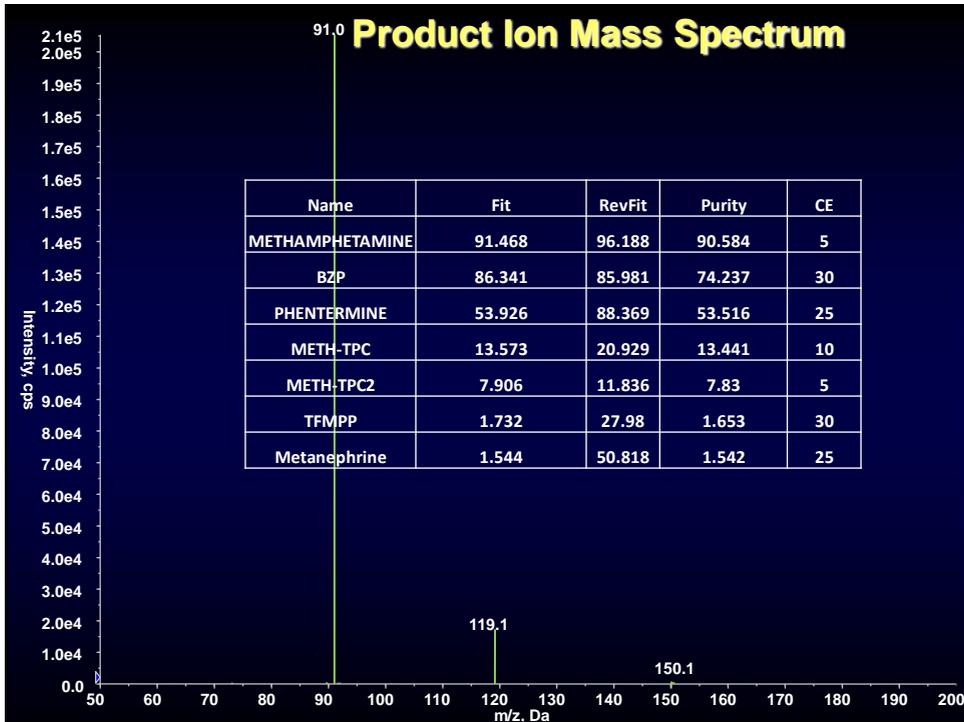
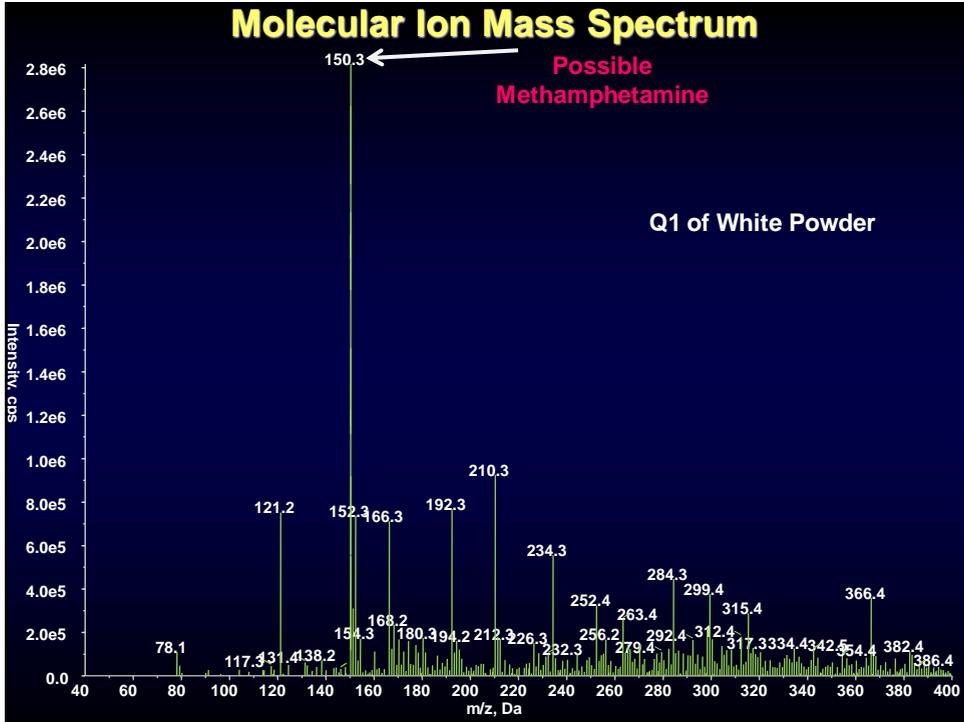


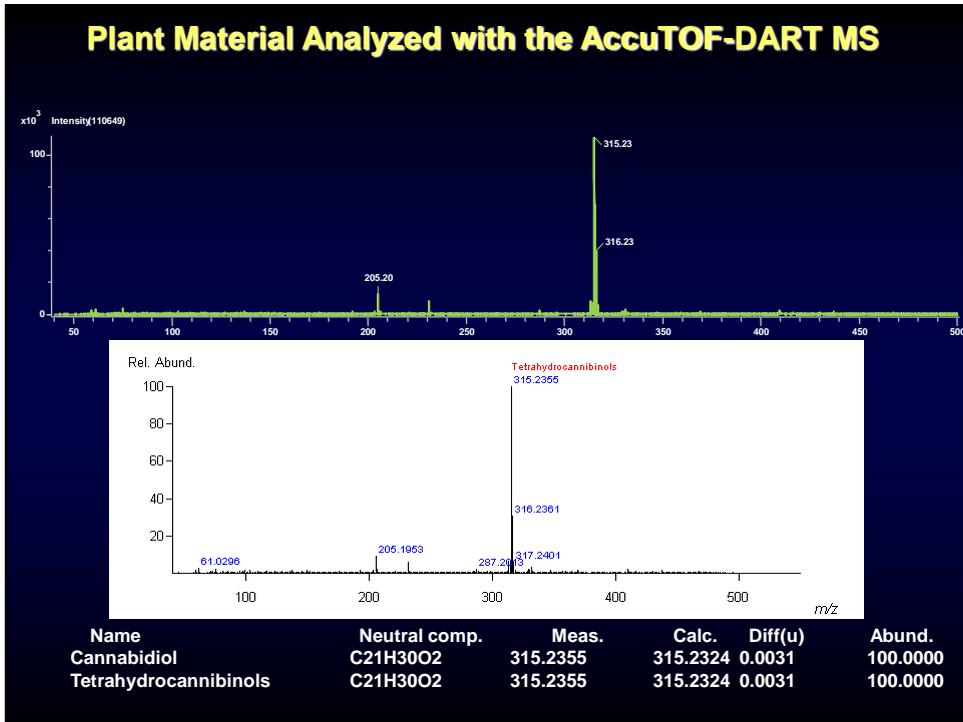
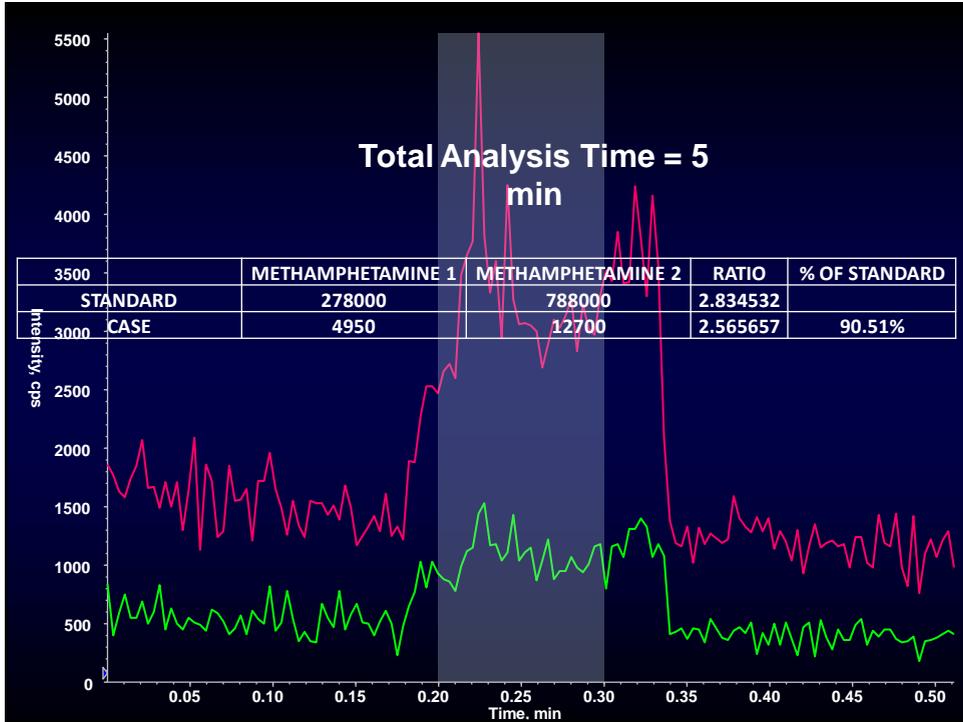
[http://www.ecstasy2.com/img/ecstasy\\_pill\\_collage1.jpg](http://www.ecstasy2.com/img/ecstasy_pill_collage1.jpg)

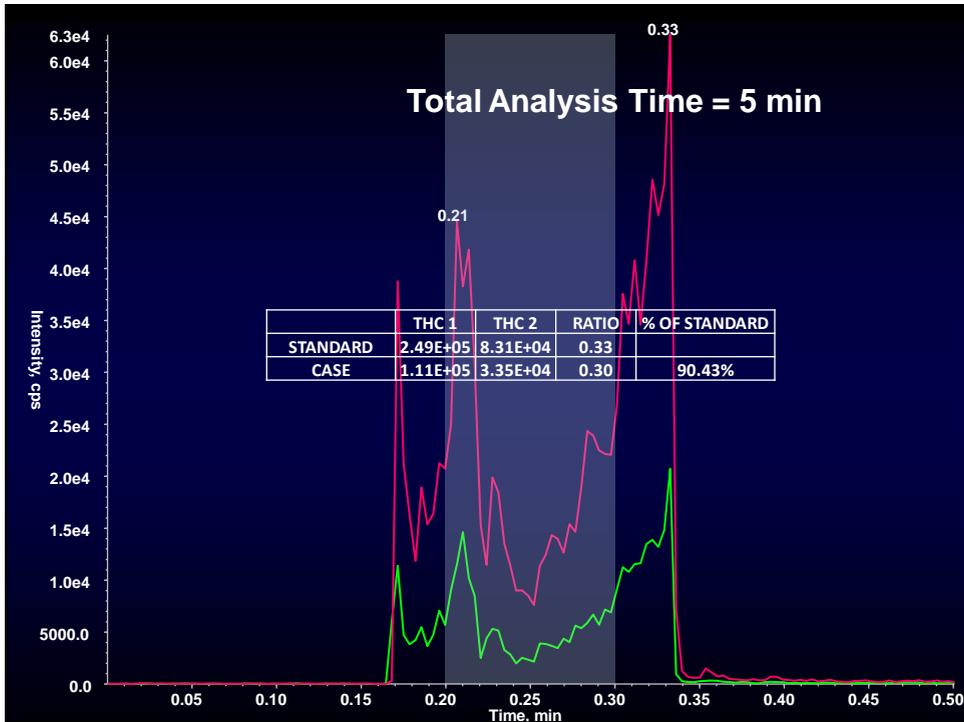
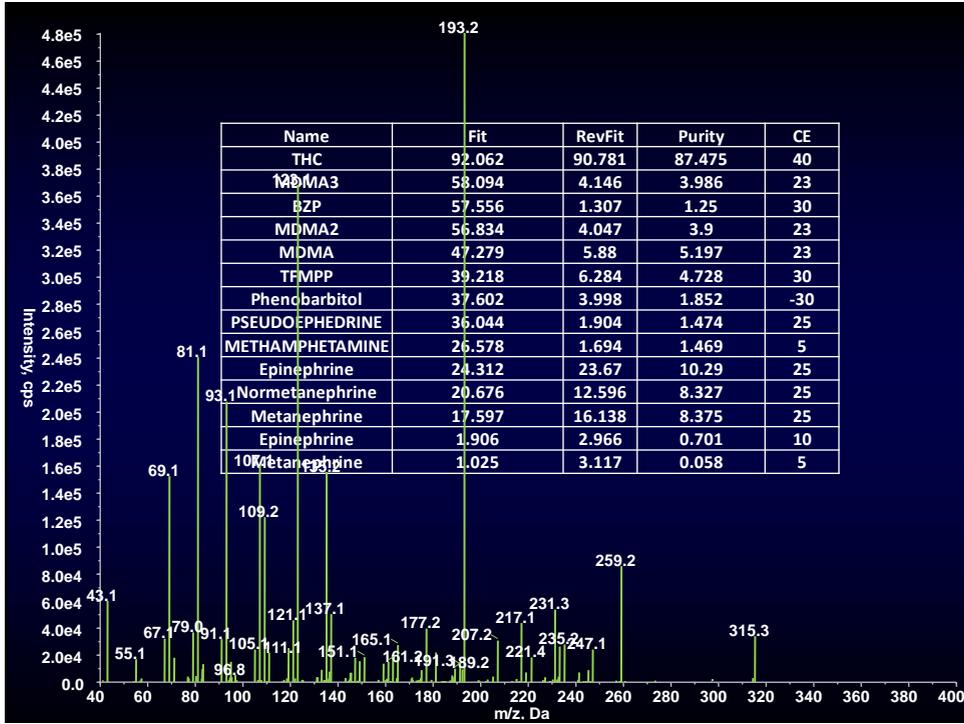


## White Powder Analyzed with the AccuTOF-DART MS

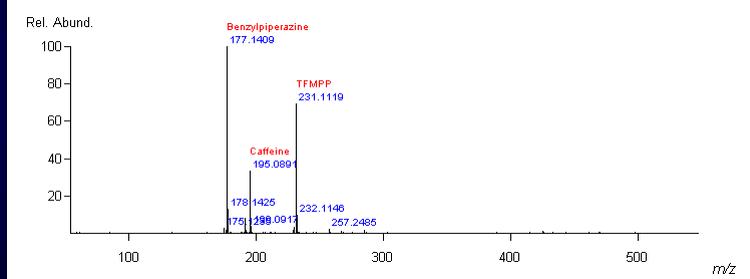
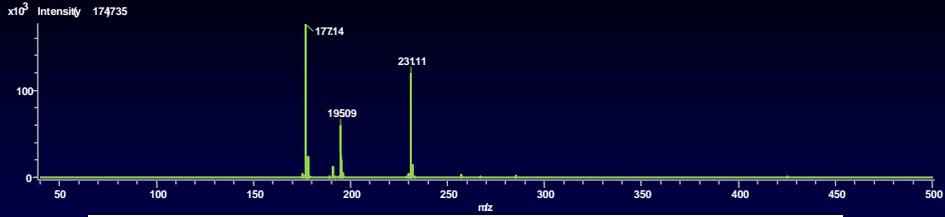




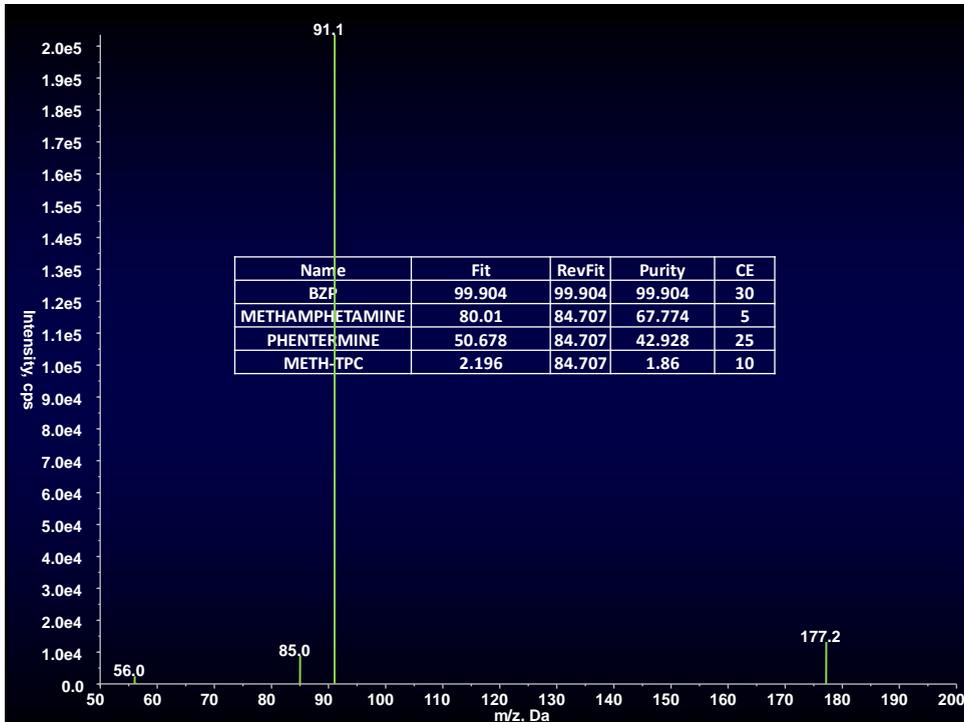


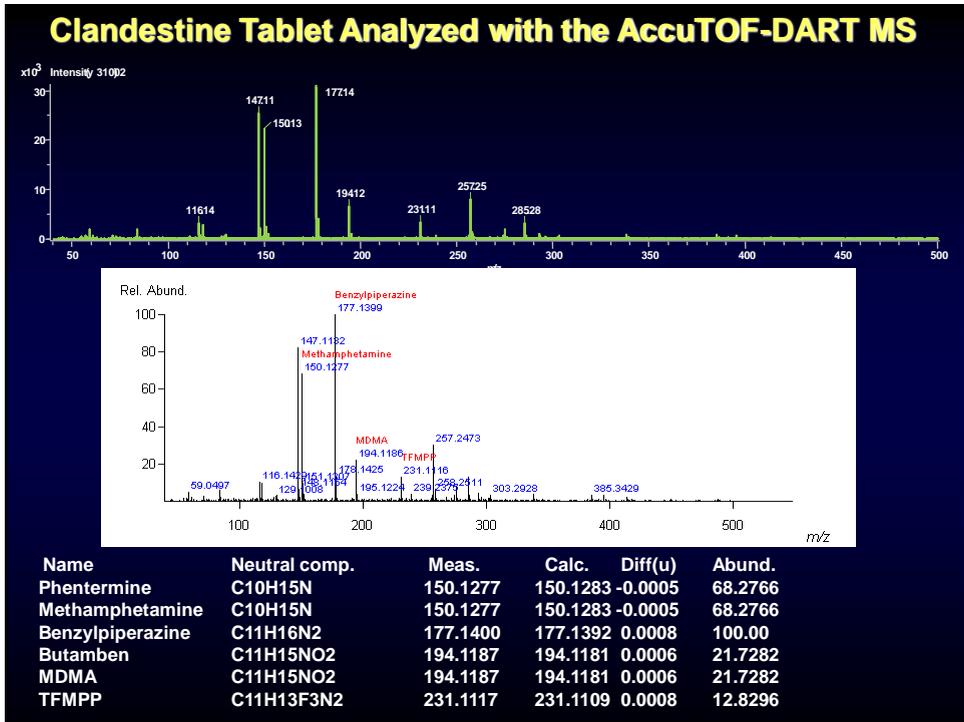
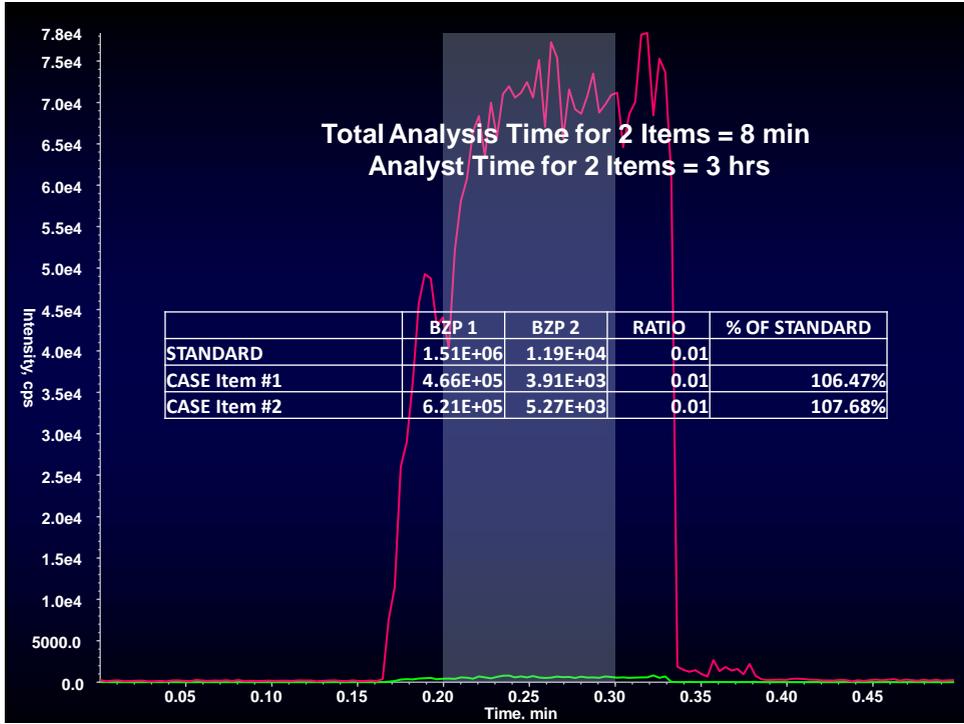


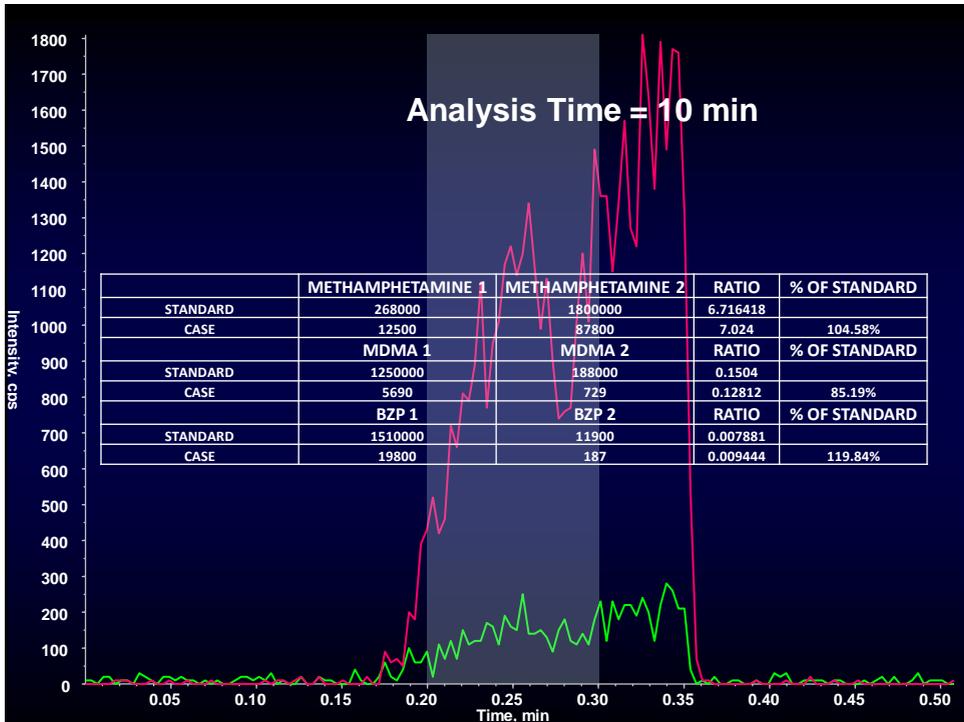
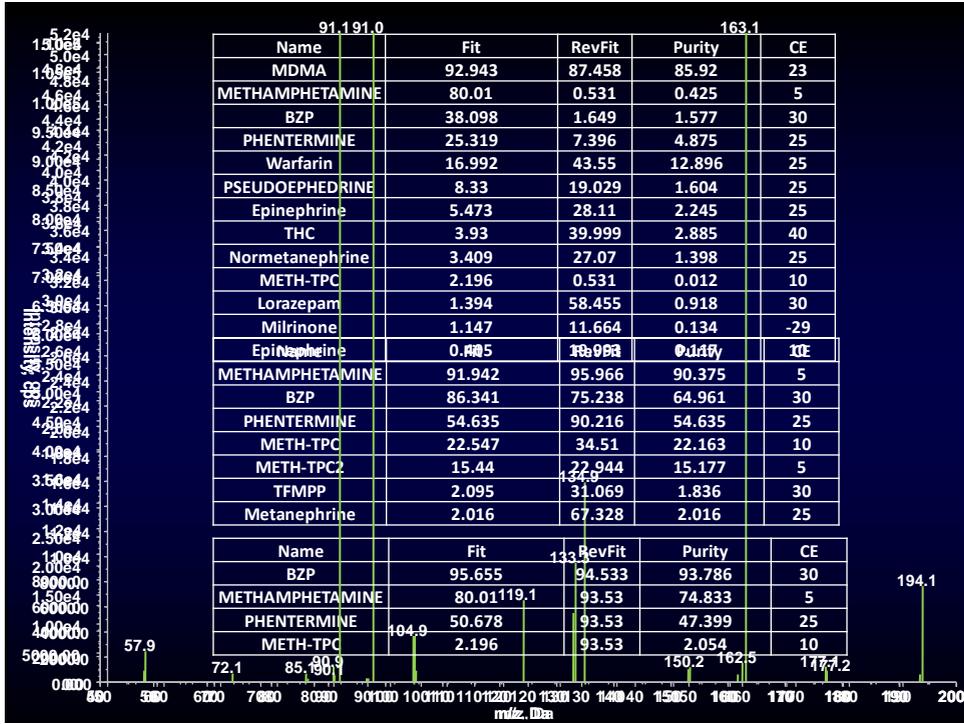
### Clandestine Tablet Analyzed with the AccuTOF-DART MS

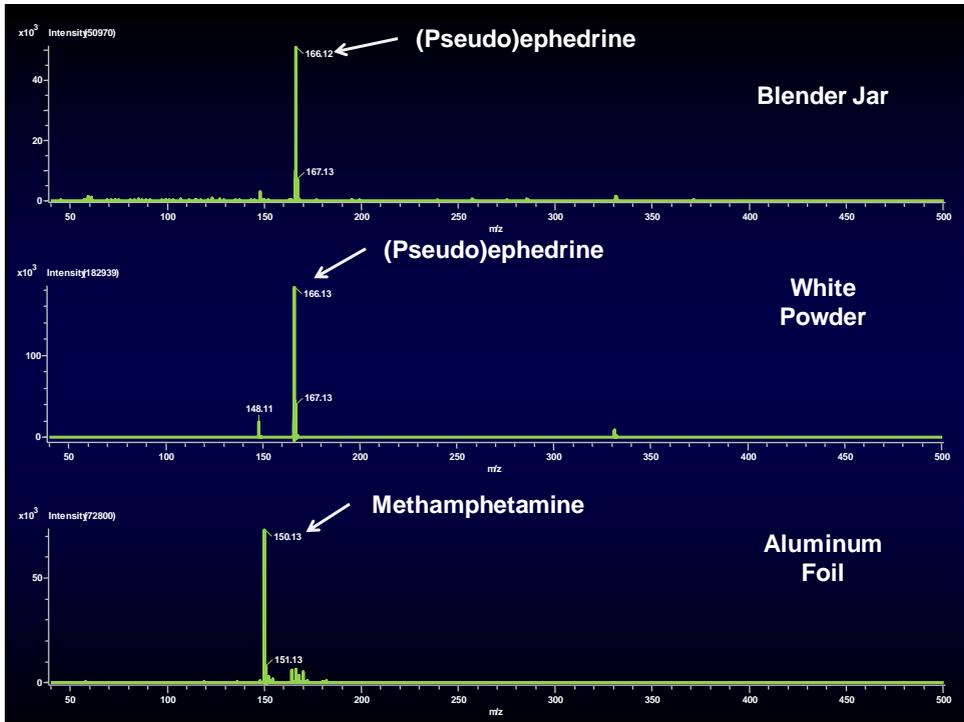
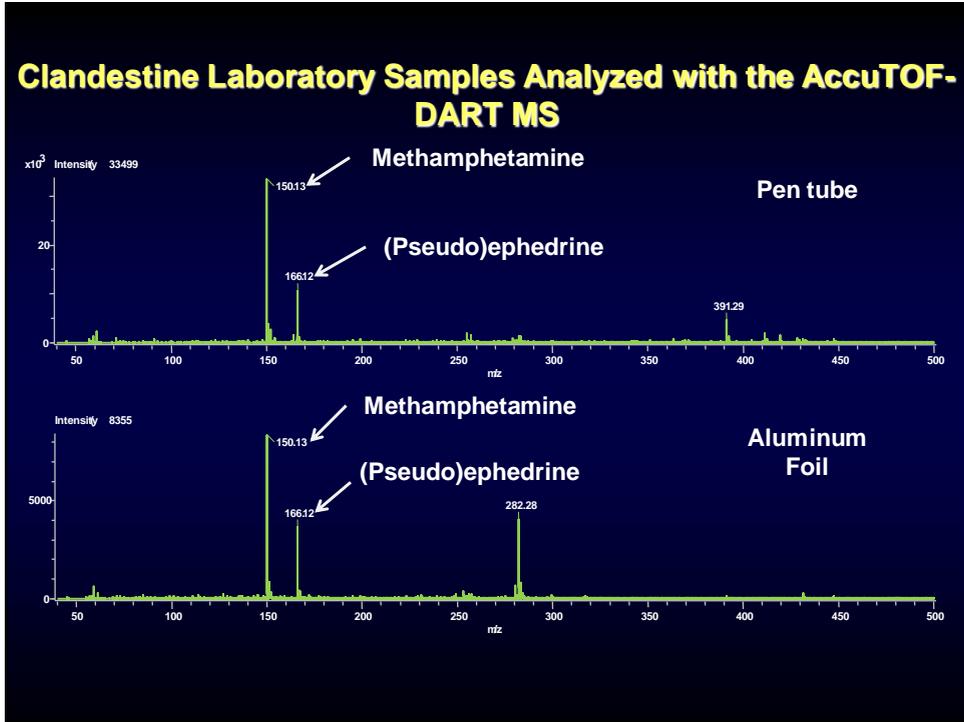


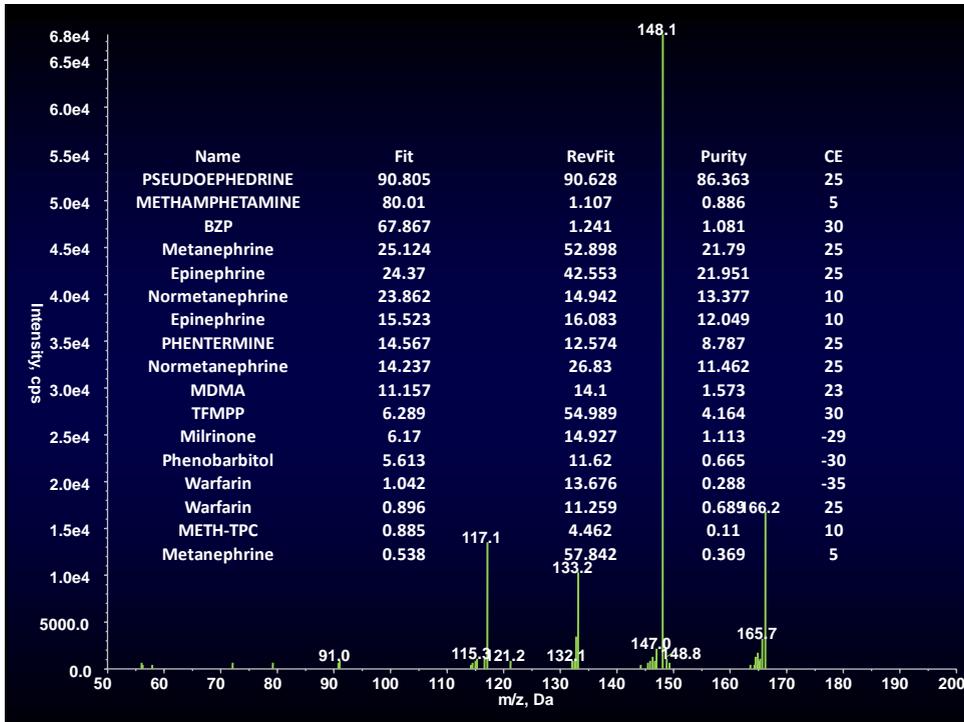
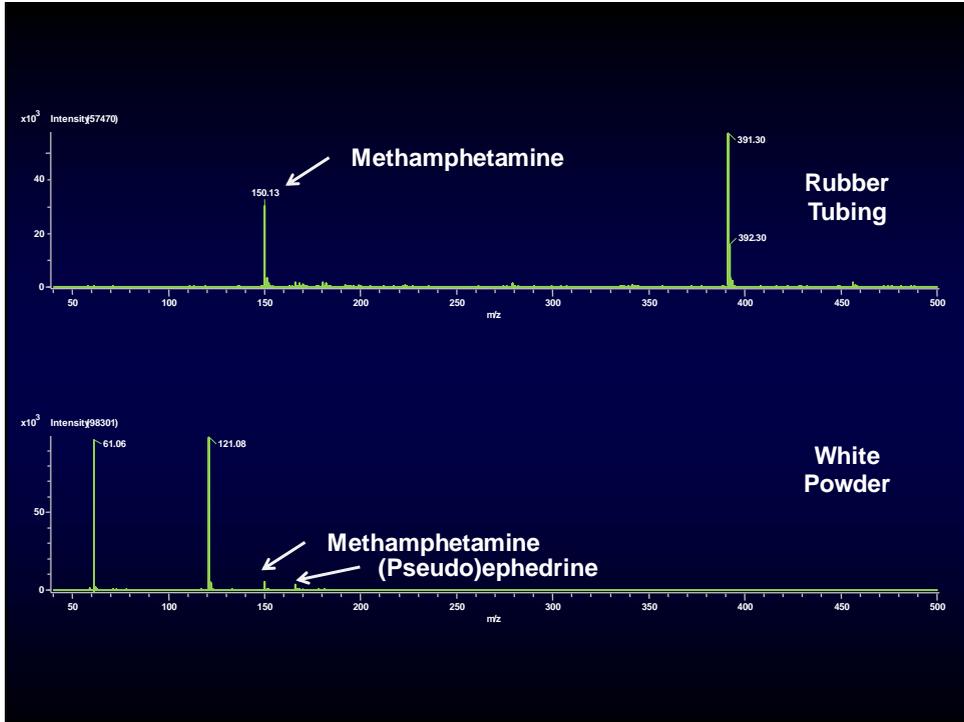
Name	Neutral comp.	Meas.	Calc.	Diff(u)	Abund.
Benzylpiperazine	C <sub>11</sub> H <sub>16</sub> N <sub>2</sub>	177.1409	177.1392	0.0017	100.0000
Caffeine	C <sub>8</sub> H <sub>10</sub> N <sub>4</sub> O <sub>2</sub>	195.0891	195.0882	0.0009	33.0937
TFMPP	C <sub>11</sub> H <sub>13</sub> F <sub>3</sub> N <sub>2</sub>	231.1119	231.1109	0.0010	69.6525











## Clandestine Laboratory Samples Analyzed with the DART-QTRAP

	METHAMPHETAMINE 1	METHAMPHETAMINE 2	RATIO	% OF STANDARD
STANDARD	2.78E+05	7.88E+05	2.83	
White Powder	1.51E+04	3.97E+04	2.63	92.75%
Pen Tube	6.85E+04	1.88E+05	2.74	96.82%
Aluminum Foil	9.25E+03	2.31E+04	2.50	88.10%
Aluminum Foil	3.02E+04	7.99E+04	2.65	93.34%
Rubber Tubing	7.89E+04	2.13E+05	2.70	95.24%

	PSEUDO 1	PSEUDO 2	RATIO	
STANDARD	5.75E+05	1.10E+05	0.19	
White Powder	5.17E+05	1.00E+05	0.19	101.11%
White Powder	4.08E+04	6.97E+03	0.17	89.30%
Pen Tube	5.75E+04	1.18E+04	0.21	107.27%
Blender	5.94E+05	1.14E+05	0.19	100.32%

Total Analysis Time for 7 Items = 1 hr  
Analyst Time for 7 Items = 2 days

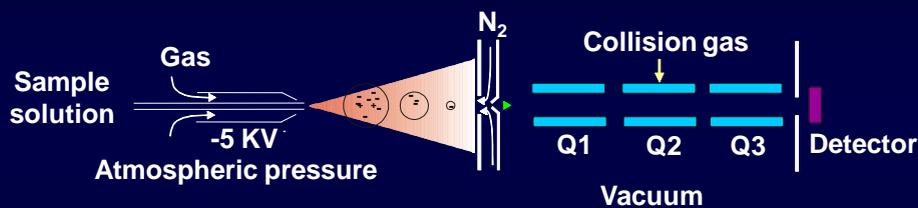
## Real Time Sample Analysis with the AccuTOF-DART MS

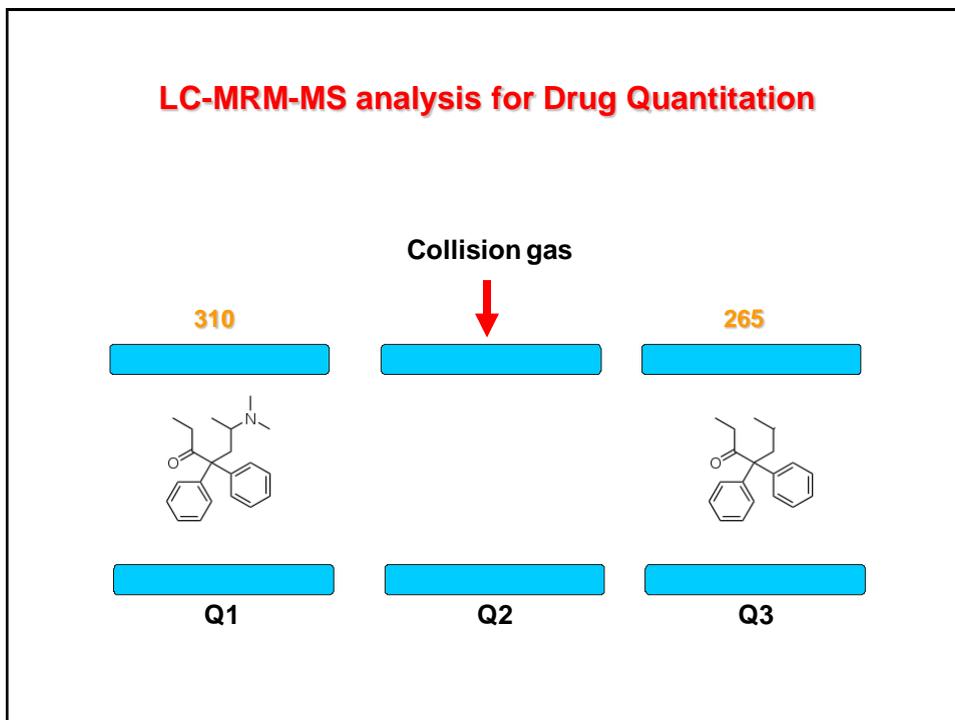
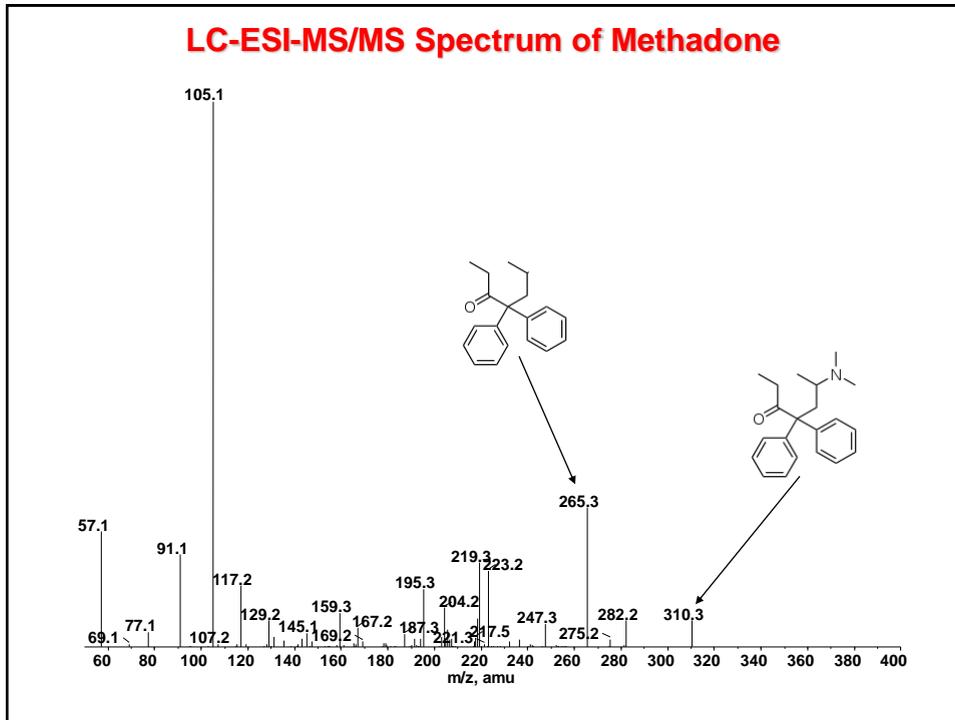
- Efficient screening instrument
  - Soft ionization keeps the molecular ion intact
    - Mass accuracy allows matches within 5 mmu of the theoretical mass of a compound
  - No extraction is required for sample analysis
    - Raw samples the preferred sample
  - High-throughput
    - Typical analysis time for a sample is 1-2 min

## Real Time Sample Analysis with the DART-QTRAP MS

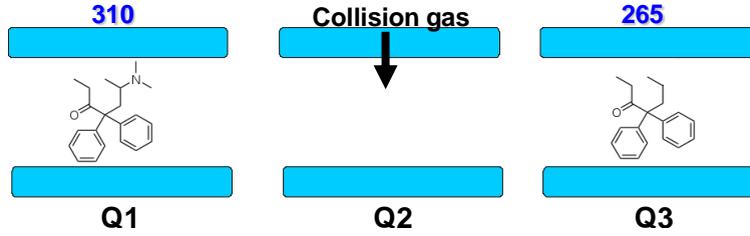
- Compound fragmentation is possible without extraction
- CID fragmentation allows retention of molecular ion in fragmentation spectrum
  - These can be searched against an in house library for identification
- MRM analysis gives ion ratios for a second level of compound identification in comparison to a standard
- Complex mixtures do not present a problem for analysis
  - The instrument has the ability to isolate a single compound for fragmentation

## LC-MRM-MS assay for Drug Detection and Quantitation



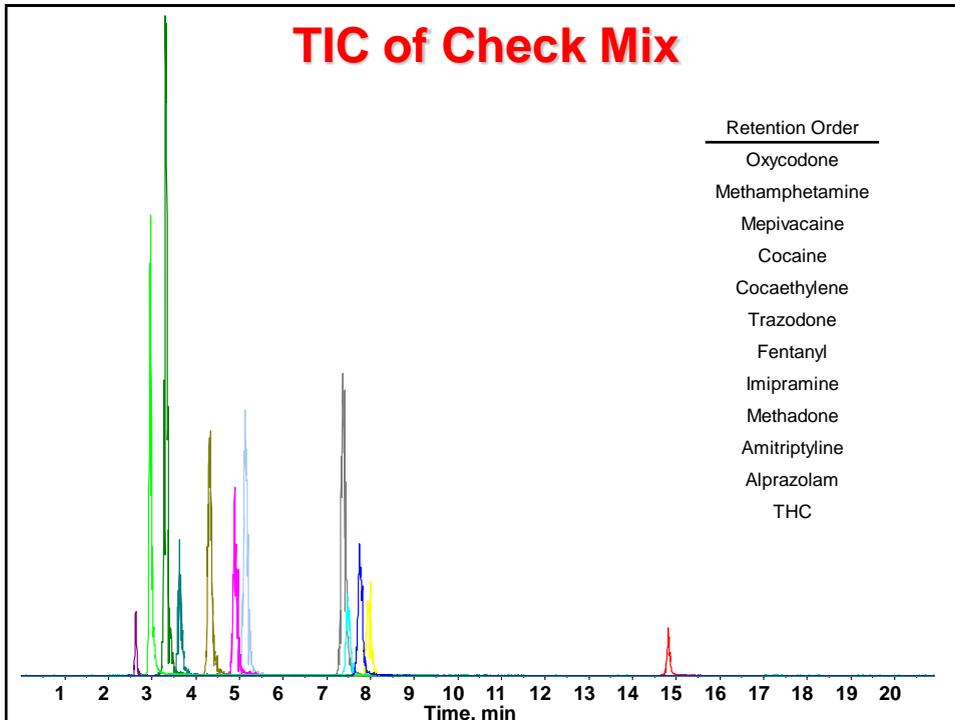


## LC-ESI-MRM-MS

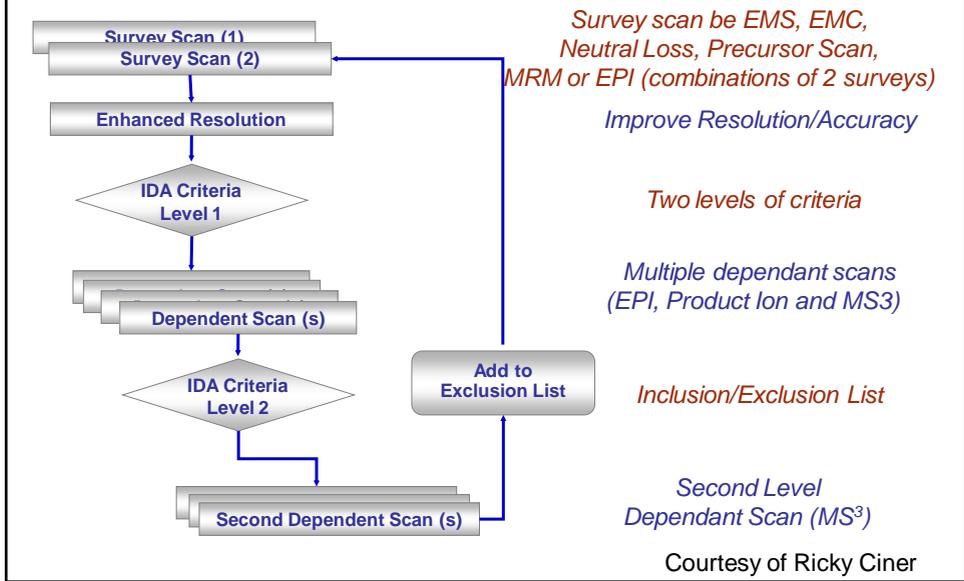


Compound	Molecular Weight	Parent ion	Product ion	Dwell Time (msec)	Declustering Potential (DP)	Collision Energy (CE)	Retention Time	pKa
Alprazolam	308.0829	309.1	205	25	60	50	8.03	2.4
Amitriptyline	277.183	278.2	91	25	45	42	7.8	9.4
Cocaehtylene	317.37	318.2	196	25	40	39	4.32	--
Cocaine	303.1471	304.1	82	25	30	40	3.6	8.6
Fentanyl	336.2202	337.2	188	25	55	43	5.18	8.4
Imipramine	280.1939	281.2	86	25	35	32	7.41	9.5
Mepivacaine	246.1732	247.2	98	25	42	28	3.32	7.6
Methadone	309.2093	310.1	265	25	30	35	7.56	8.6
Methamphetamine	149.1204	150.1	91	25	34	27	2.96	8.6
Oxycodone	315.1471	316.1	241	25	50	40	2.6	8.5
THC	314.2246	315.1	193	25	37	34	14.91	10.6
Trazodone	371.1513	372.2	176	25	60	42	4.87	6.1

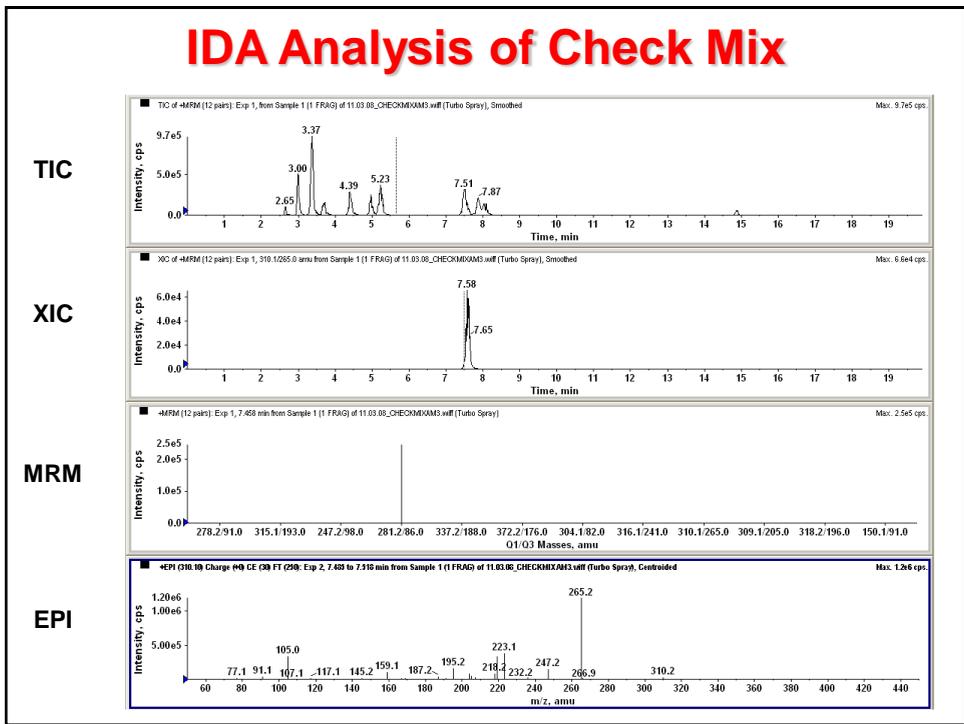
## TIC of Check Mix

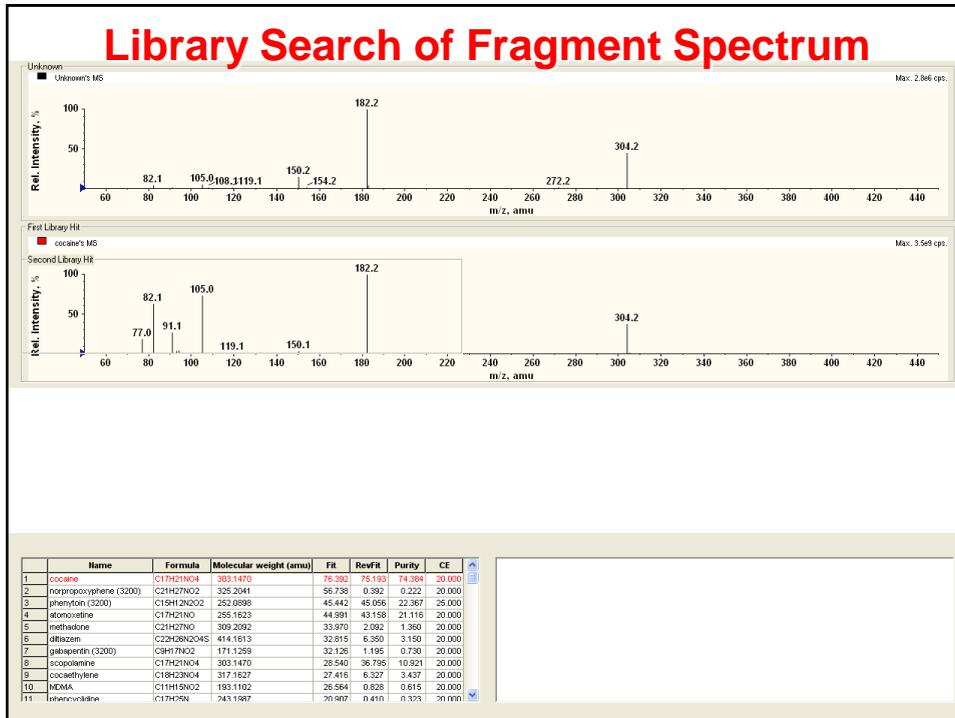


# Information Dependent Acquisition (IDA)



## IDA Analysis of Check Mix





## Summary

- LC-ESI-MS can be used in the qualitative and quantitative analysis of drugs in toxicological specimens
  - The instrumentation is advantageous in that chemicals do not have to be derivatized
  - The soft ionization aids detection of the parent ion of the compound

## Overall Summary

- Mass spectrometry is a powerful tool in a forensic science lab
- New instrumentation is expanding the sample analysis possibilities beyond current limitations
- No one technique is robust enough for everything, therefore a combination of techniques is ideal for screening and confirmation of drug and toxicology samples

## Acknowledgements

### UAB

- Dr. Stephen Barnes
- Marion Kirk
- Ray Moore
- Dr. Matthew Renfrow
- Landon Wilson

### ADFS

- Dr. Dale Carpenter
- Andrea Headrick
- Dr. Jack Kalin
- Gary Wallace