

Mass Spectrometry based metabolomics

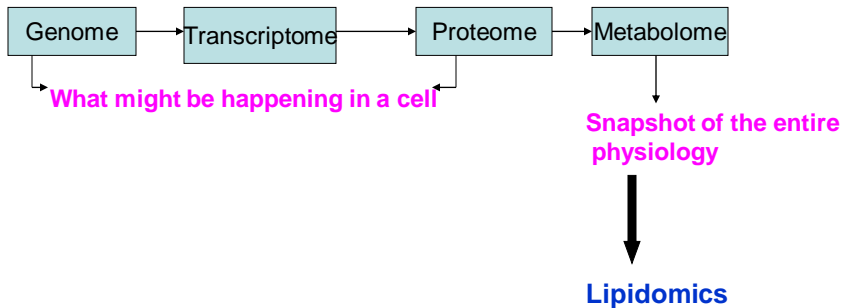
Metabolomics- A realm of small molecules (<1000 Da)

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What is metabolomics?

- **Identification and quantification of the complete set of metabolites in a biological system**
- **Quantitative global analysis of metabolites from cells, tissues and fluids**
- **Quantitative measurement of the dynamic metabolite response of living systems to pathophysiological stimuli or genetic modification**

Metabolomics in the context of other omics



Lipids are important- as a membrane bilayer

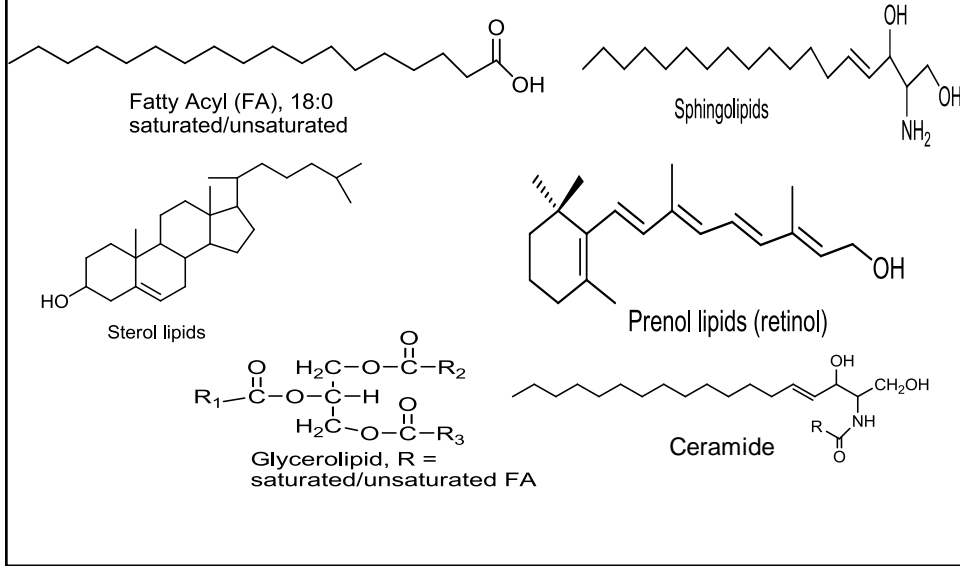
- provides hydrophobic environment for protein function
- reservoir of energy
- signaling molecules

Lipidomics can perhaps best be defined as a comprehensive analysis of lipids on the systems-level scale together with their interacting factors

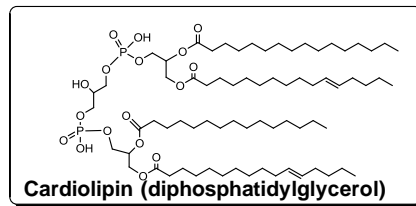
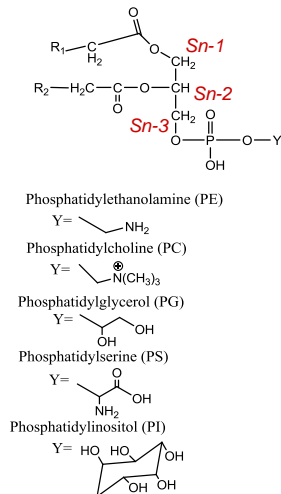
Outlines

- **Brief introduction to lipidomics**
- **Analytical methodology: MS/MS structure elucidation of phospholipids**
- **Phospholipid analysis in lean and ob/ob mice by mass spectrometry**
- **MS/MS analysis of eicosanoids**

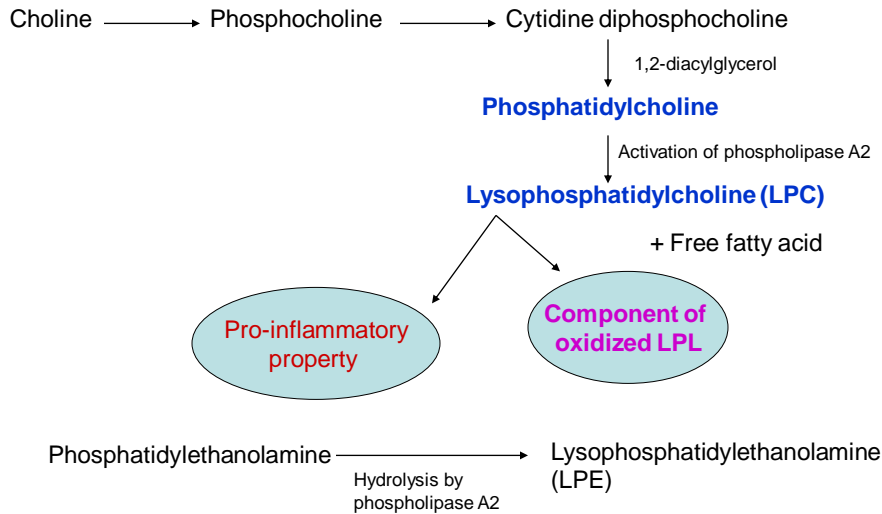
Structures of different lipids classes



Structures of main phospholipids



How phospholipids are synthesized?



Extraction of lipids by Bligh/Dyer method

- To a homogenized sample (1 ml containing internal standards) add methanol (2.5 ml) and chloroform (1.25 ml), sonicate by 4-5 bursts and added 1.0 ml water and 1.25 ml chloroform additionally and vigorously shaken.
- Centrifuge (1,000 x g) for 2 min and separate the chloroform layer (bottom layer) and repeat the process twice.
- Combine the chloroform soluble phase and evaporate to dryness and stored at -20 °C until analysis.

Shotgun lipidomics: intrasource separation of lipids for quantitative lipidomics

Group	Electrical Propensity	Lipid Classes
Anionic lipids	Carry net negative charge(s) at physiological pH	Cardiolipin, acylCoA, sulfatide, PtdIns (PtdInsP, PtdInsP ₂ , PtdInsP ₃), PtdGro, PtdSer, PtdH, etc.
Weak anionic lipids	Carry a net negative charge at alkaline pH	PE, lysoPE, ceramide, NEFA, eicosanoids, etc.
Neutral polar lipids	Neutral at alkaline pH	PC, lysoPC, SM, glycolipid, TAG, etc.
Special lipids	Vary	Acylcarnitine, sterols, etc.

The ionization efficiency of an analyte greatly depends on the electrical propensity of an individual analyte in its own microenvironment to lose or gain a charge

Source: Gross and Han,, 2004

Which ionization mode for which phospholipids?

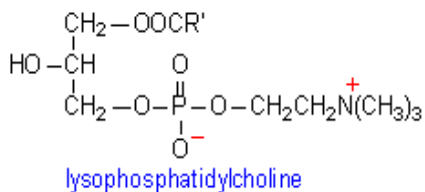
Positive ion mode

PC
LPC
PE
LPE
SM
PS

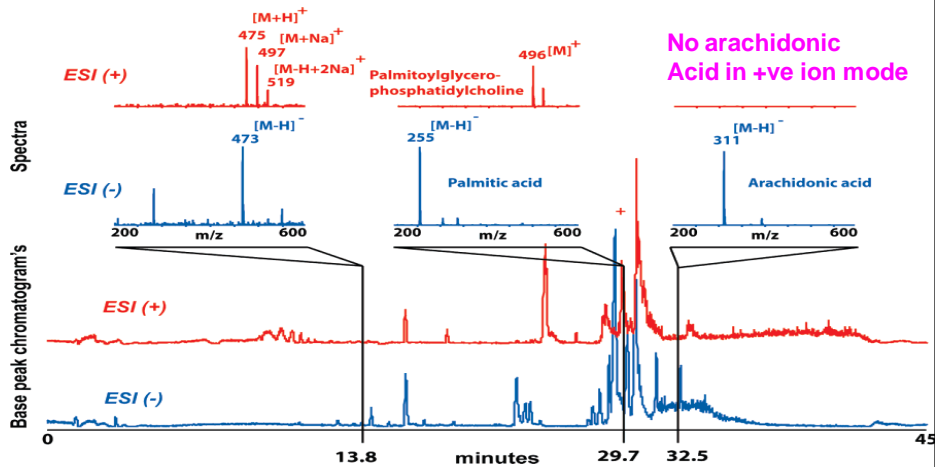
Negative ion mode

PE
PA
PI
PI
PG
PIPs

PC = phosphatidylcholine
PA = phosphatic acid
PE = phosphatidylethanolamine
PS = phosphatidylserine
PG = phosphatidylglycerol
PI = phosphatidylinositol
PIP = PI monophosphate
SM = sphingomyelin
LPE = lysoPE



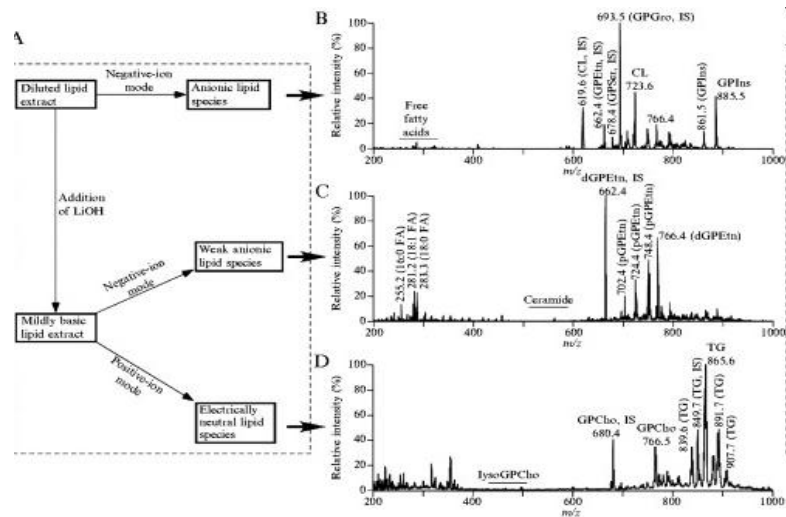
Increasing metabolite coverage using +ve and -ve ion mode



Representative Q1 scans of a methanolic extract of human blood serum

Source: Nordstrom et al. Analytical Chemistry, 2007

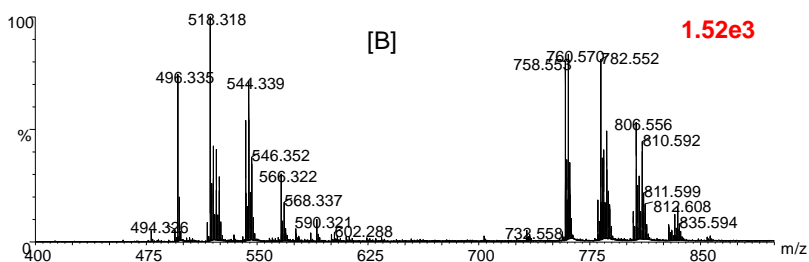
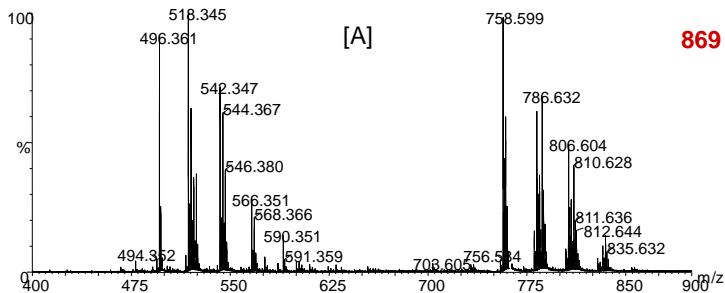
Application of shotgun lipidomics: intra-source separation of lipids



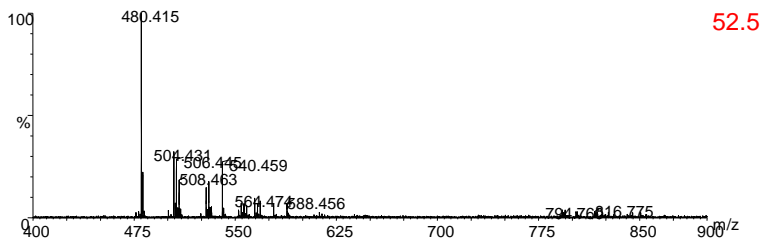
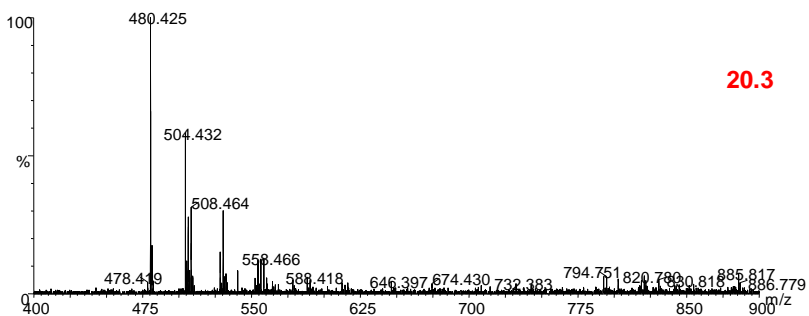
Source: Gross and Han, methods in Enzymology, 2007

Total scan of metabolites (Q1 SCAN + ion mode) for a plasma sample obtained from lean mouse [A]; ob/ob mouse

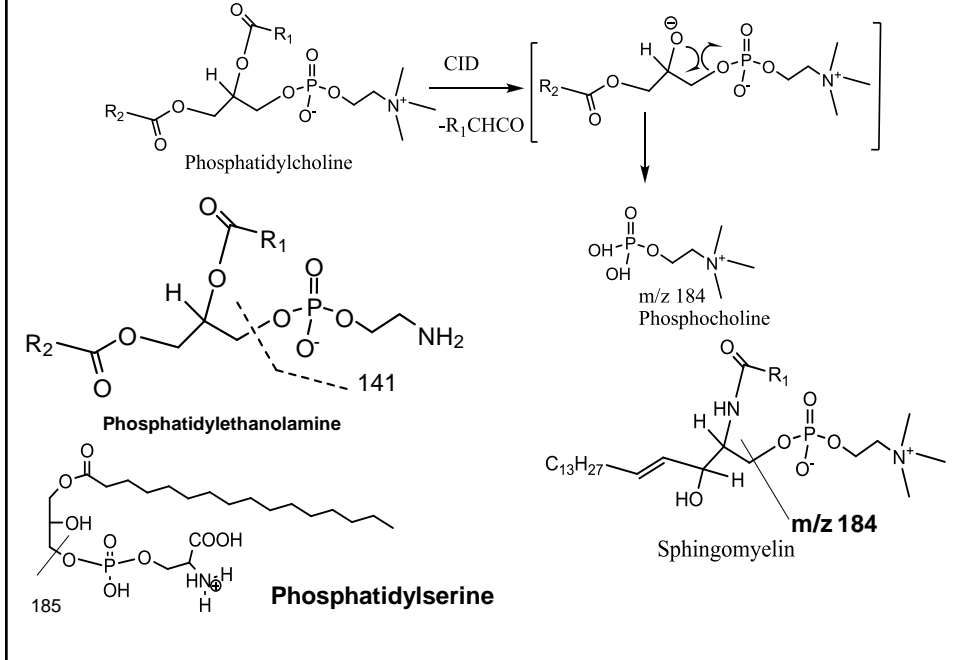
Total metabolomics



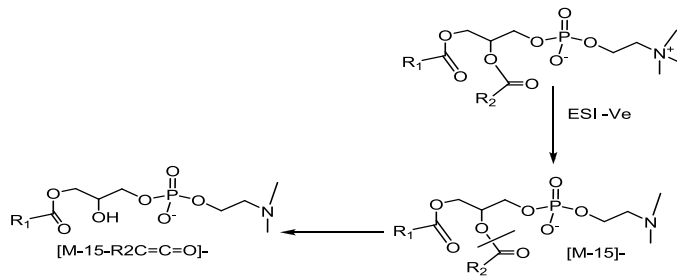
Total scan of metabolites (Q1 SCAN -ve ion mode) for a plasma sample obtained from lean mouse [A]; ob/ob mouse



Tandem mass spectrometry (MS/MS) of phospholipids



MS/MS in negative ion mode of phospholipids provide information about the fatty acyl chain



ESI-MS/MS analyses of various lipids

Lipid Class(s)	Precursor Ion	MS/MS Mode & Conditions	Fragment
cardiolipin	[M-2H] ²⁻	PI, <i>m/z</i> 153.0, 35 eV	glycerol phosphate derivative
PtdGro, PtdH	[M-H] ⁻	PI, <i>m/z</i> 153.0, 35 eV, *	glycerol phosphate derivative
PtdIns	[M-H] ⁻	PI, <i>m/z</i> 241.1, 45 eV	cyclic inositol phosphate
		PI, <i>m/z</i> 153.0, 35 eV	glycerol phosphate derivative
PtdInsP	[M-H] ⁻	PI, <i>m/z</i> 321.1, 53 eV	phosphoinositol phosphate
PtdInsP ₂	[M-H] ⁻	PI, <i>m/z</i> 401.1, 62 eV	diphosphoinositol phosphate
PtdSer	[M-H] ⁻	NL, 87.0 amu, 25 eV, *	serine
		PI, <i>m/z</i> 153.0, 35 eV	glycerol phosphate derivative
sulfatide	[M-H] ⁻	PI, <i>m/z</i> 97.0, 65 eV	sulfate
acylCoA	[M-2H] ²⁻	PI, <i>m/z</i> 339.0, 30 eV, *	doubly-charged CoA derivative
PE, lysoPE	[M-H] ⁻	PI, <i>m/z</i> 196.0, 50 eV	glycerol phosphoethanolamine derivative
ceramide	[M-H] ⁻	NL, 256.2 amu, 32 eV *	
		NL, 327.3 amu, 32 eV	
		NL, 240.2 amu, 32 eV *	2- <i>trans</i> -palmitoyl alcohol
PC, lysoPC, SM	[M+Li(Na)] ⁺	NL, 59.1 amu, -28 eV, *	trimethylamine
	[M+Li(Na)] ⁺	NL, 183.1 amu, -32 eV	phosphocholine
	[M+Li] ⁺	NL, 189.1 amu, -42 eV	lithium cholinephosphate
	[M+Na] ⁺	NL, 205.1 amu, -35 eV	sodium cholinephosphate
	[M+H] ⁺	PI, <i>m/z</i> 184.1, -30 eV, *	phosphocholine
	[M+Cl] ⁻	NL, 50.0 amu, 24 eV, *	methylchloride
cerebroside	[M+Li] ⁺	NL, 162.2, -50 eV, *	
	[M+Cl] ⁻	NL, 36.0 amu, 30 eV	hydrogen chloride
MGDG	[M+Li(Na)] ⁺	PI, <i>m/z</i> 227(243), -45 eV	Li(Na)+galactose derivative
DGDG	[M+Li(Na)] ⁺	PI, <i>m/z</i> 227(243), -66 eV	Li(Na)+galactose derivative
acylcarnitine	[M+H] ⁺	PI, <i>m/z</i> 85.1, -20 eV, *	carnitine
chol. ester	[M+NH ₄] ⁺	PI, <i>m/z</i> 369.3, -50 eV, *	cholestane cation
TAG	[M+Li] ⁺	NL, X amu, -35 eV	a fatty acid

Source: Gross and Han,, 2004

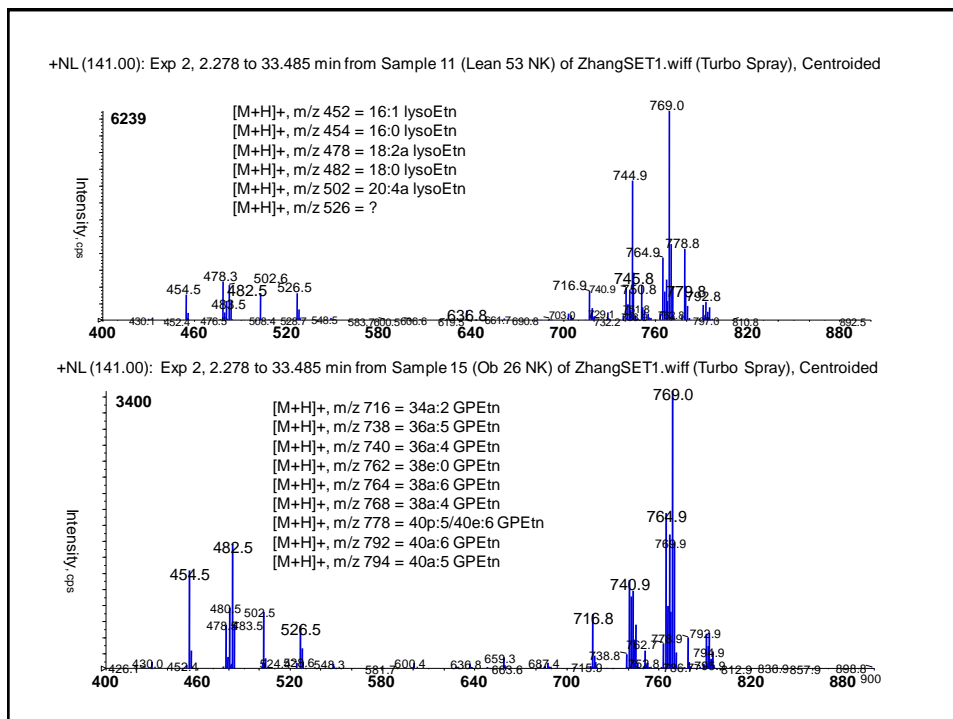
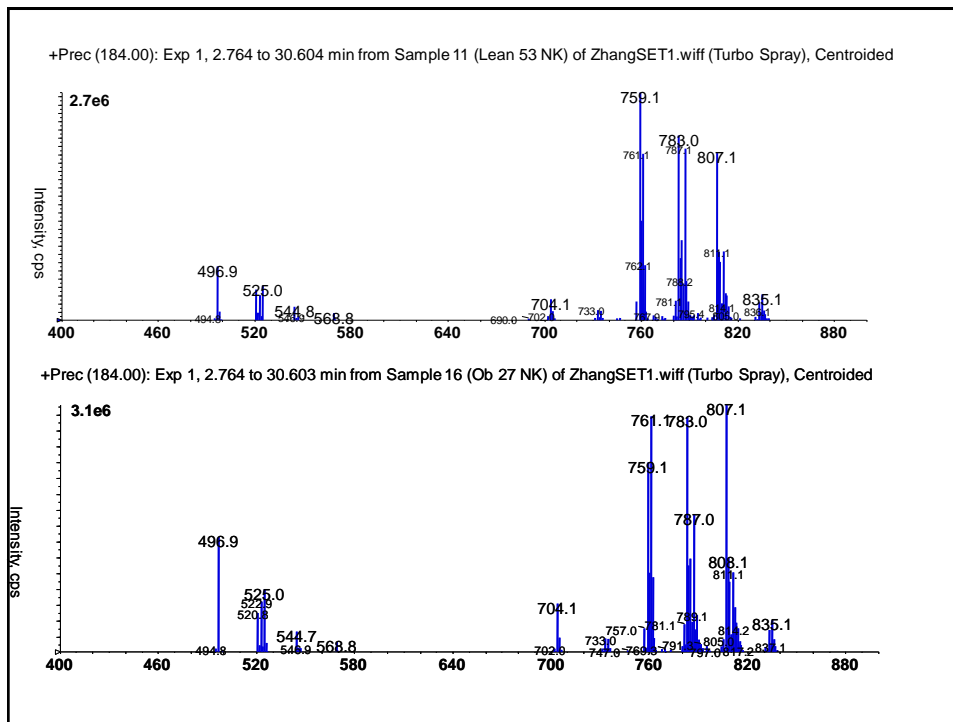
Focused lipidomics

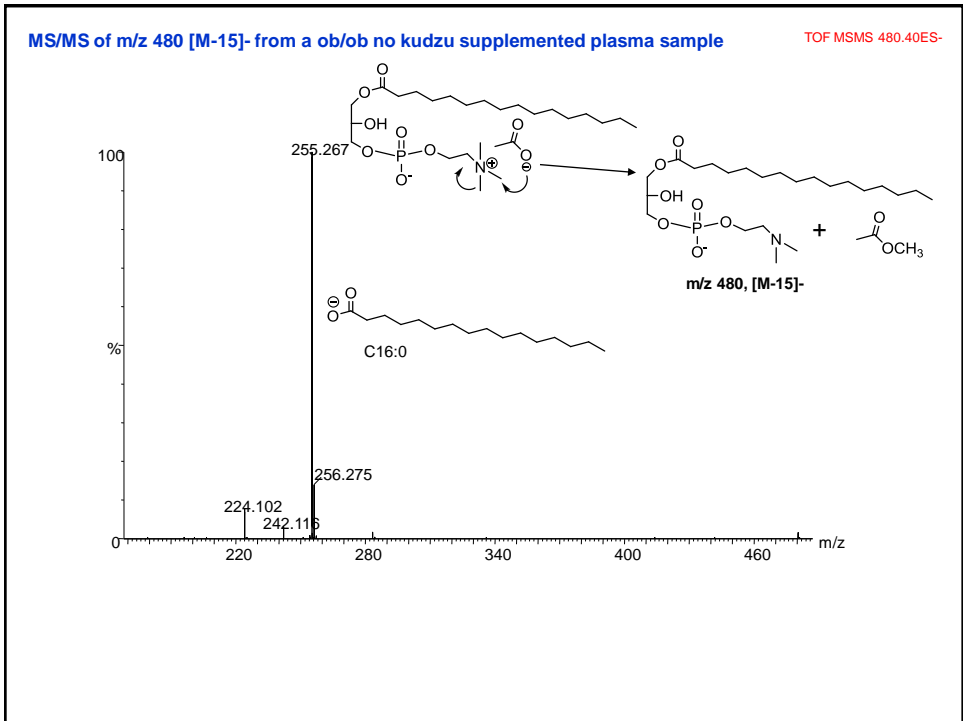
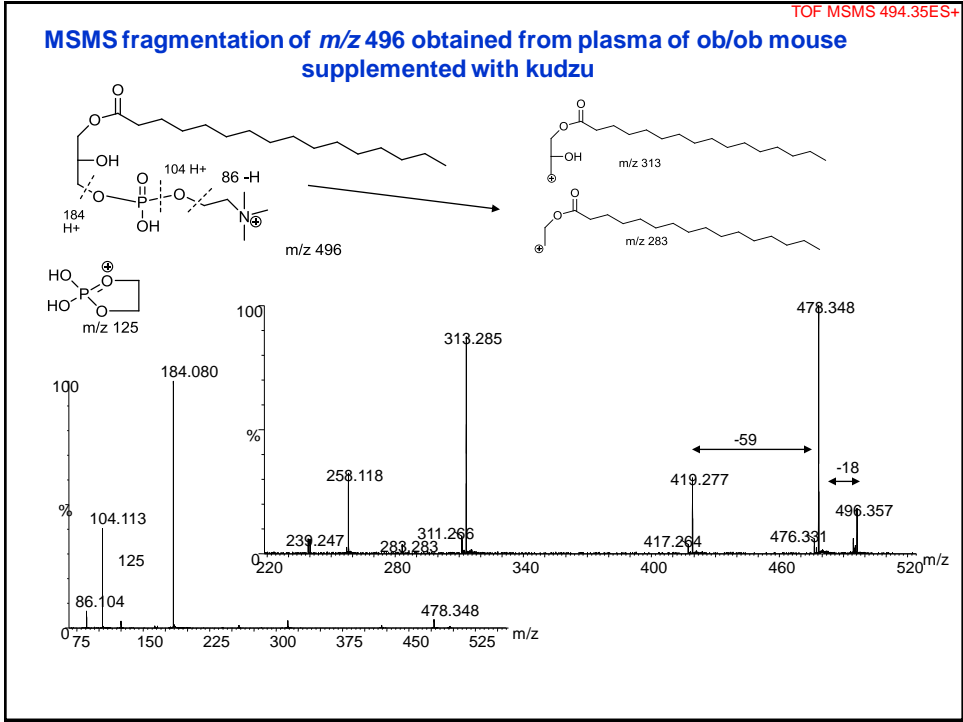
A. Flow injection (ESI-MS/MS)

-Precursor ion scanning at *m/z* 184-choline-containing phospholipids +ve ion mode

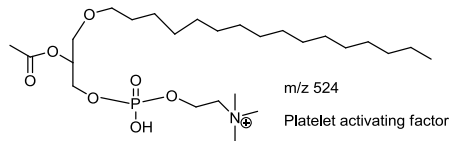
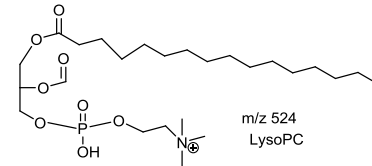
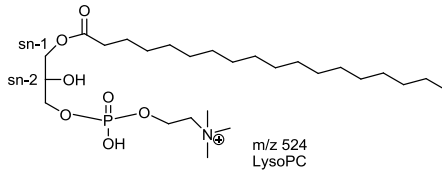
- Neutral scanning of 141, 185, 189, and 277 u used for PE, PS, phosphatidylglycerol (PG), and phosphatidylinositol (PI), respectively

- precursor ion scanning at *m/z* 153 and 241 in -ve ion mode-glycerol-containing phospholipids and inositol-containing phospholipids, respectively

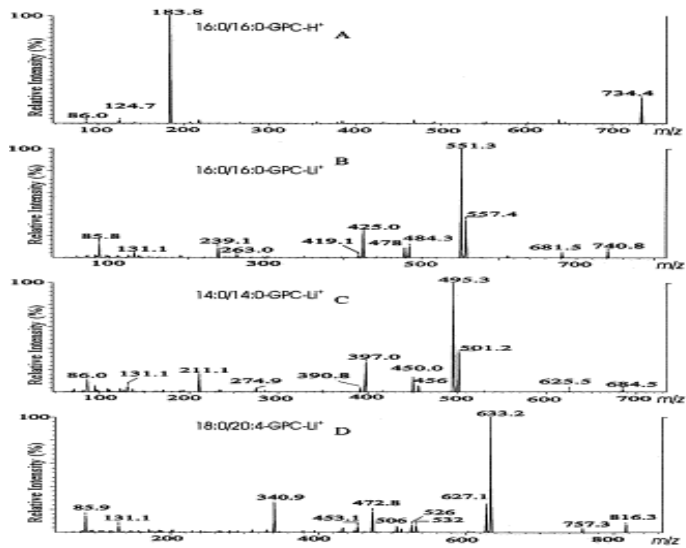




Several isomeric compounds exists and unambiguous identification is a challenge

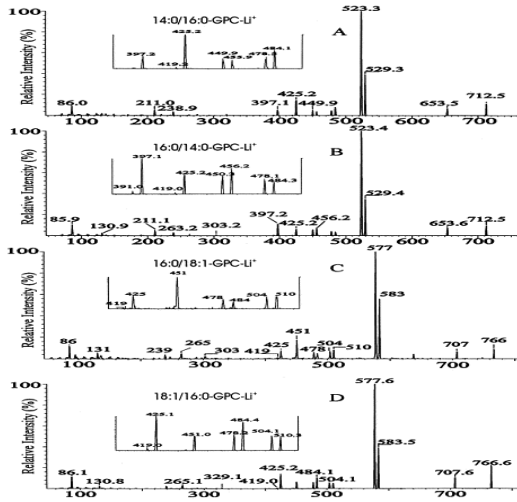


Lithiated adducts of phosphocholine provide more structural information in their MS/MS spectra



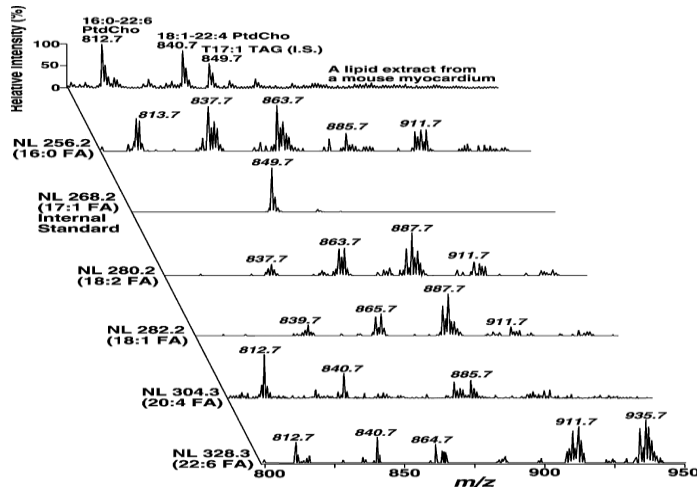
Source: Hsu et al. J. Am Soc. Mass Spectrom, 1998

Relative abundances of product ion can be used to distinguish positional isomers of lithiated phospholipids



Source: Hsu et al. J. Am Soc. Mass Spectrom, 1998

A 2D ESI mass spectrometric finger print for TG molecules



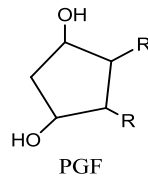
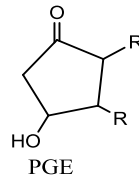
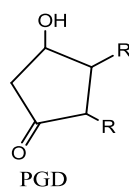
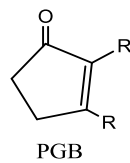
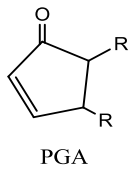
Source: Han and Gross, 2004

MS/MS analysis of eicosanoids

Eicosanoids, meaning 20
derived from a 20-carbon acid,
arachidonic acid

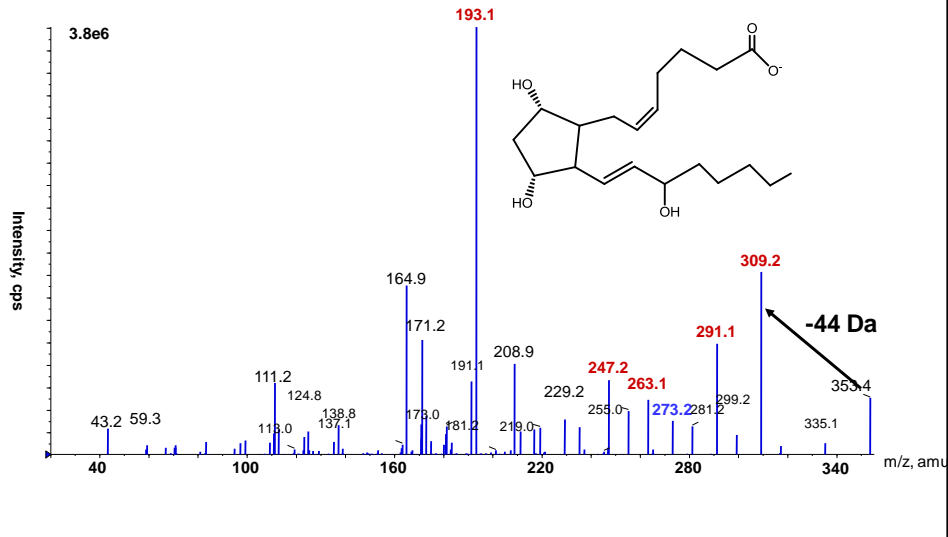
-Important lipid mediators and elicit potent effects in various biological systems mediated through specific protein receptors

Structural representation PG based on ring features

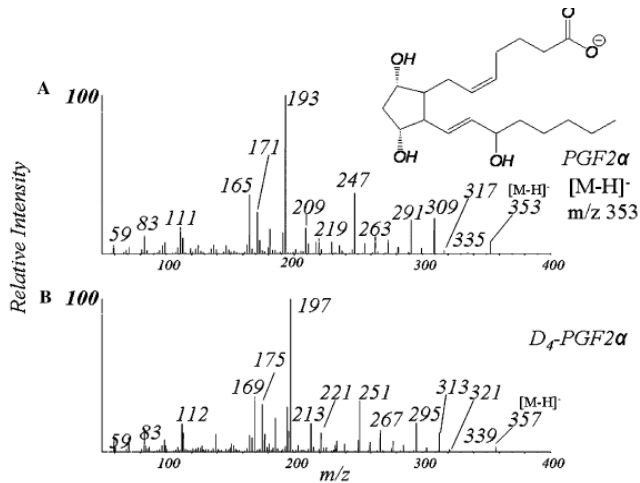


R = aliphatic chain

ESI-MS/MS of the [M-H]⁻ from PGF₂ α m/z 353 using a quadrupole mass spectrometer

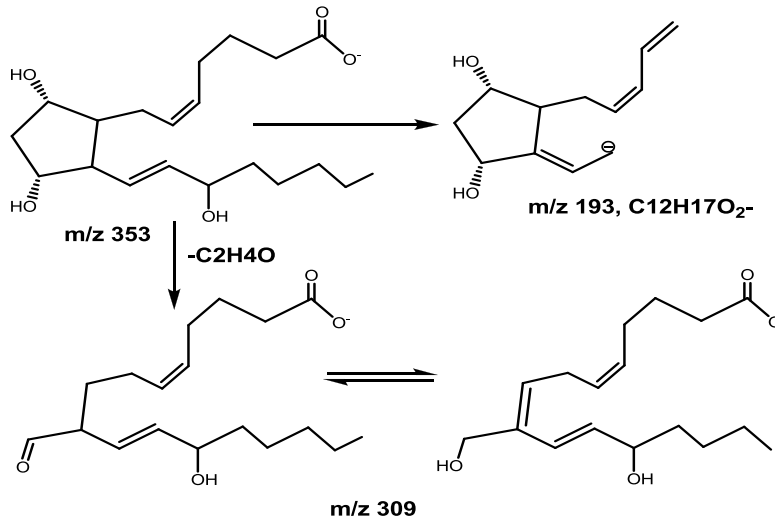


What information does deuterium labeling at C-2 and C-3 of PGF₂ provide us for structure elucidation of PG?



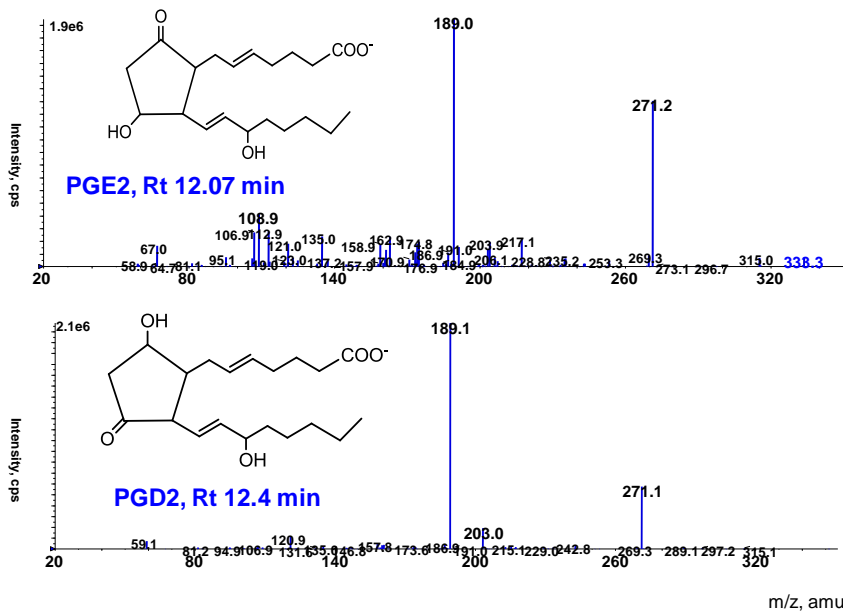
Source: Murphy et al. Analytical Biochemistry, 2005

Fragmentation scheme of PGF2 α [M-H]⁻ m/z 353

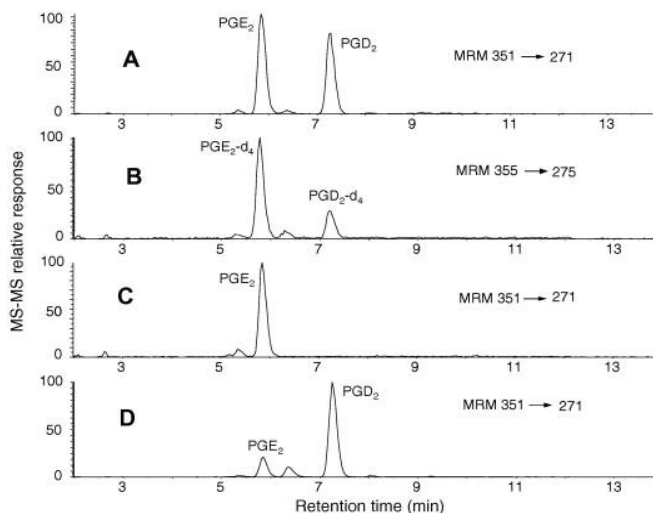


Ions m/z 309, 291, 273 and 193 are indicative of F2-ring

MS/MS fragmentation of PGE2 and PGD2 m/z 351.00



Deuterated PG standards are used for quantitative analysis of PGs in a extract



Source: Cao et al. Analytical Biochemistry, 2008

Library search for eicosanoid <http://www.lipidmaps.org/>

LIPID MAPS -- LIPID Metabolites And Pathways Strategy

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LMSPD: Lipid classification search results

Fatty Acyls [FA] (W) --> Eicosanoids [FA03]

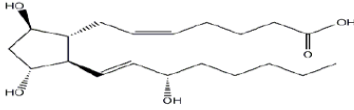
LM_ID	Common Name	Systematic Name	Formula	Mass
LMFA03000001	8(9)-EpETE	(+/-)-8(9)-epoxy-5Z,11Z,14Z,17Z-eicosatetraenoic acid	C ₂₀ H ₃₀ O ₃	318.22
LMFA03000002	11(12)-EpETE	(+/-)-11(12)-epoxy-5Z,8Z,14Z,17Z-eicosatetraenoic acid	C ₂₀ H ₃₀ O ₃	318.22
LMFA03000003	14(15)-EpETE	(+/-)-14(15)-epoxy-5Z,8Z,11Z,17Z-eicosatetraenoic acid	C ₂₀ H ₃₀ O ₃	318.22
LMFA03000004	17(18)-EpETE	(+/-)-17(18)-epoxy-5Z,8Z,11Z,14Z-eicosatetraenoic acid	C ₂₀ H ₃₀ O ₃	318.22
LMFA03000005	11(R)-HEDE	11R-hydroxy-12E,14Z-eicosadienoic acid	C ₂₀ H ₃₆ O ₃	324.27
LMFA03000006	17R,18S-EpETE	17R,18S-epoxy-5Z,8Z,11Z,14Z-eicosatetraenoic acid	C ₂₀ H ₃₀ O ₃	318.22
LMFA03000008	15(R)-HEDE	15R-hydroxy-11Z-13E-eicosadienoic acid	C ₂₀ H ₃₆ O ₃	324.27
LMFA03000009	11S-HEDE	11S-hydroxy-12E,14Z-eicosadienoic acid	C ₂₀ H ₃₆ O ₃	324.27
LMFA03010000	Prostanoid acid skeleton	-	-	-
LMFA03010001	6-keto-PGF1α	6-oxo-9S,11R,15S-trihydroxy-13E-prostanoic acid	C ₂₀ H ₃₄ O ₆	370.24
LMFA03010002	PGF2α	9S,11R,15S-trihydroxy-5Z,13E-prostadienoic acid	C ₂₀ H ₃₄ O ₅	354.24
LMFA03010003	PGE2 (W)	9-oxo-11R,15S-dihydroxy-5Z,13E-prostadienoic acid	C ₂₀ H ₃₂ O ₆	352.22
LMFA03010004	PGD2 (W)	9S,15S-dihydroxy-11-oxo-5Z,13E-prostadienoic acid	C ₂₀ H ₃₂ O ₆	352.22
LMFA03010005	PGA1	9-oxo-15S-hydroxy-10Z,13E-prostadienoic acid	C ₂₀ H ₃₂ O ₄	336.23
LMFA03010006	PGF2α-d4	9S,11R,15S-trihydroxy-5Z,13E-prostadienoic acid (3,3,4,4-d4)	C ₂₀ H ₃₀ O ₄ O ₅	356.27
LMFA03010007	PGD2-d4	9S,15S-dihydroxy-11-oxo-5Z,13E-prostadienoic acid (3,3,4,4-d4)	C ₂₀ H ₂₈ O ₄ O ₅	356.25
LMFA03010008	PGE2-d4	11R,15S-dihydroxy-9-oxo-5Z,13E-prostadienoic acid (3,3,4,4-d4)	C ₂₀ H ₂₈ O ₄ O ₅	356.25
LMFA03010009	PGG2	9S,11R-epidoxo-15S-hydroperoxy-5Z,13E-prostadienoic acid	C ₂₀ H ₃₂ O ₆	368.22

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LIPID Metabolites And Pathways Strategy

Structure database (LMSD)

LMFA03010025



LM ID	LMFA03010025
Common Name	PGF ₂ β
Systematic Name	9R,11R,15S-trihydroxy-5Z,13E-prostadienoic acid
Synonyms	-
Exact Mass	354.24
Formula	C ₂₀ H ₃₄ O ₅
Category	Fatty Acyls [FA]
Main Class	Eicosanoids [FA03]
Sub Class	Prostaglandins [FA0301]
LIPIDBANK ID	XPB1764
PubChem Substance ID (SID)	4265968
KEGG ID	-

Conclusions

- **Shotgun lipidomics approaches are high throughput and applicable to perform profiling as well as quantitative analysis of various lipids in biological samples.**
- **Tandem mass spectrometry analysis of phospholipids in +ve ion mode characterizes phospholipid polar head groups, whereas –ve ion mode provide fatty acid chain structural information**
- **Identification of phospholipids at a molecular level present a great challenge due to their structural diversity and dynamic metabolism.**