

BMG/PHR 744
02-02-11 class

Nucleic acid and nucleotide mass spectrometry

Stephen Barnes, PhD

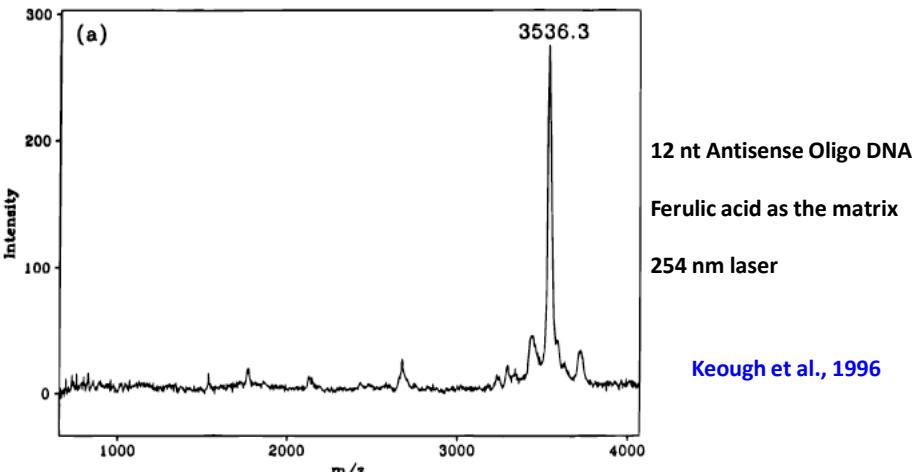
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Synopsis

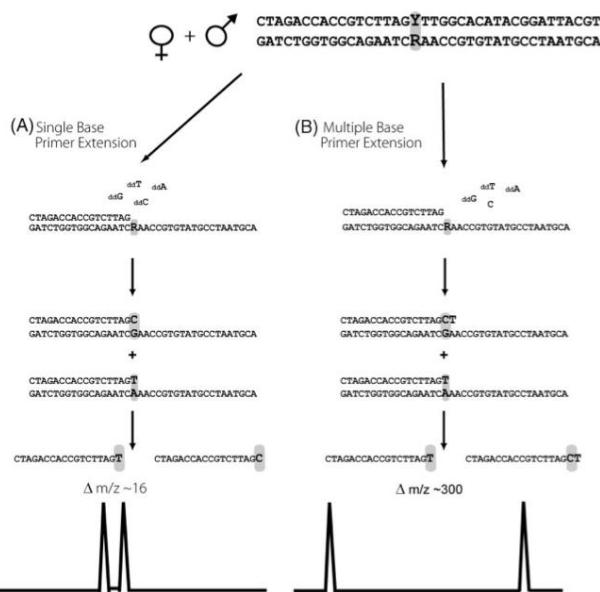
- **MALDI**
 - Sample prep
 - UV laser
 - SNPs, multiplexing
 - IR laser (large DNA/RNA)
- **ESI**
 - Bases
 - Nucleosides
 - Nucleotides

UV-MALDI TOF of nucleic acids

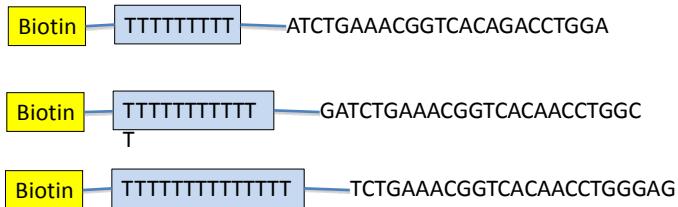
- Limited in the number of residues (up to 70 nt)



Single nucleotide polymorphisms (SNPs)

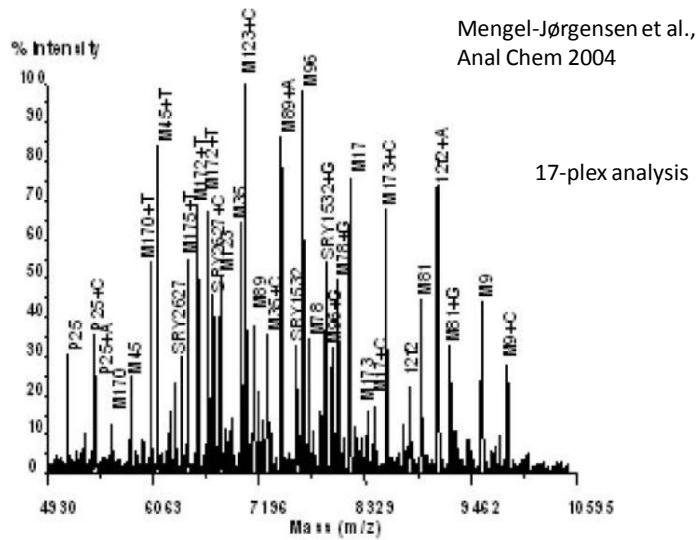


Multiplexing concept for SNPs



Δ masses						
Adenine 135.13	Adenosine	251.242	-	24.025	-15.999	- 9.013
Cytosine 111.10	Cytidine	227.217	-24.025	-	-40.024	-15.012
Guanine 151.13	Deoxyguanosine	267.241	15.999	40.024	-	25.012
Thymine 126.11	Deoxythymidine	242.229	- 9.013	15.012	-25.012	-

Multiplexed MALDI-TOF spectra



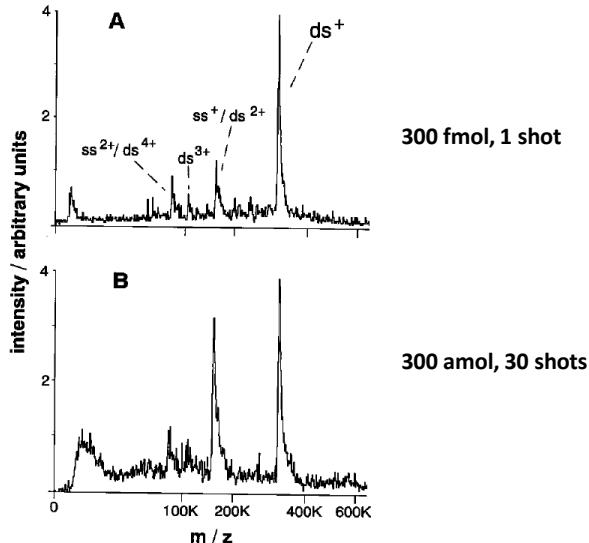
Infra-red MALDI-TOF of DNA

516 nt PCR product

Pptd with EtOH from 2 M ammonium acetate

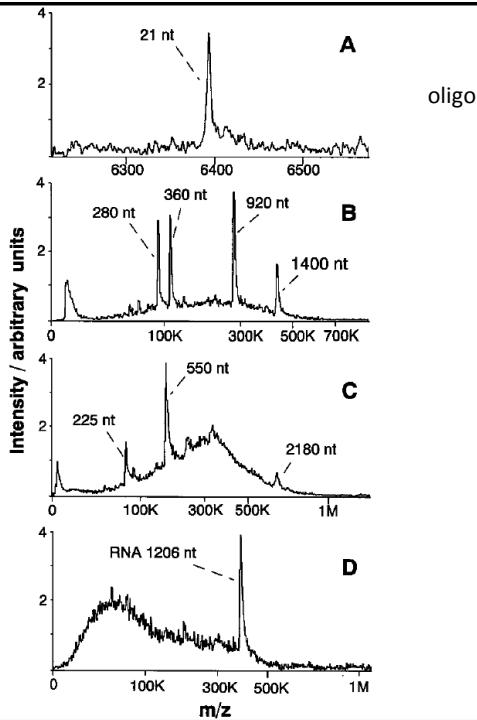
IR laser at 2.4 μ m

Sample in glycerol



Berkenkamp et al., Science 1998

IR MALDI of Nucleic acids



Pptd with EtOH from 2 M ammonium acetate

IR laser at 2.4 μ m

Sample in glycerol

Berkenkamp et al., Science 1998

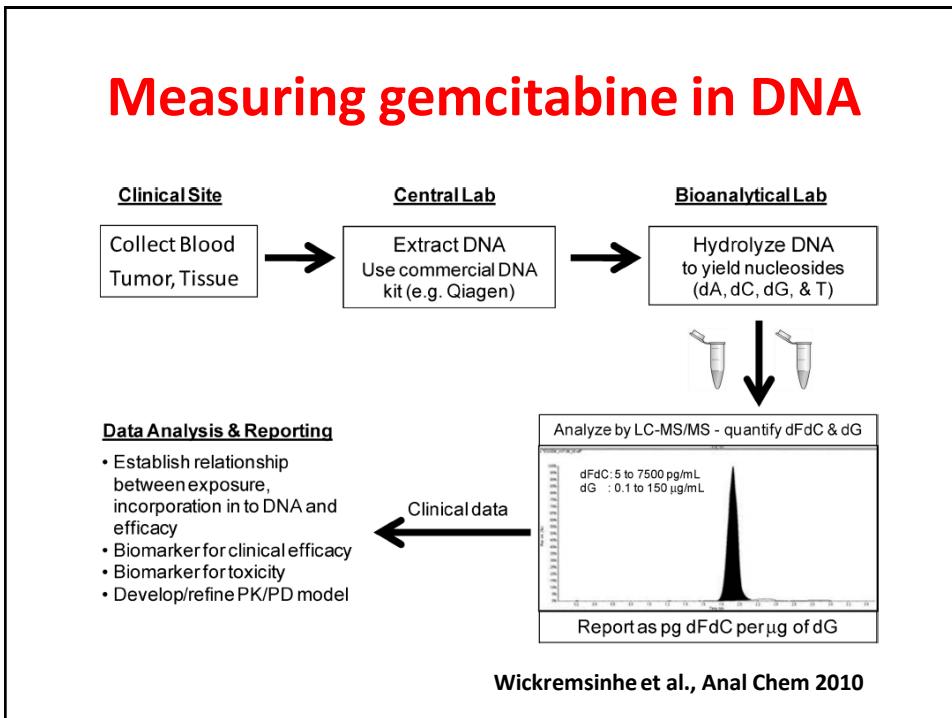
Fundamentals of nucleic acids analysis by ESI-mass spectrometry

- Analysis as bases, nucleosides
 - No different from other small molecules
 - Bases are sufficiently hydrophobic to be analyzed by reverse-phase LC-MS
 - The addition of ribose or deoxyribose not a problem
 - Detected in the positive ion mode
- Nucleotides are more difficult
 - The phosphate group reduces net charge
 - Na^+ salts cause multiple ion states

LC-MS analysis of nucleosides

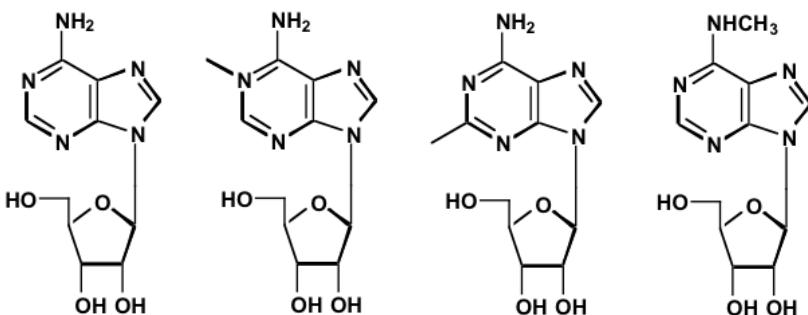
- There are nucleoside drugs
 - e.g., gemcitabine (2,2'-diflurocytidine)
 - Xu et al., J. Chromatogr B (2004)
 - LC-MS – negative ESI – m/z 262
 - Honeywell et al., J Chromatogr B (2007)
 - LC-MRM-MS – positive APCI – m/z 264/112
 - Bowen et al., J Chromatogr B (2009)
 - uPLC-MRM-MS – dansyl derivatives – 1.5 min

Measuring gemcitabine in DNA

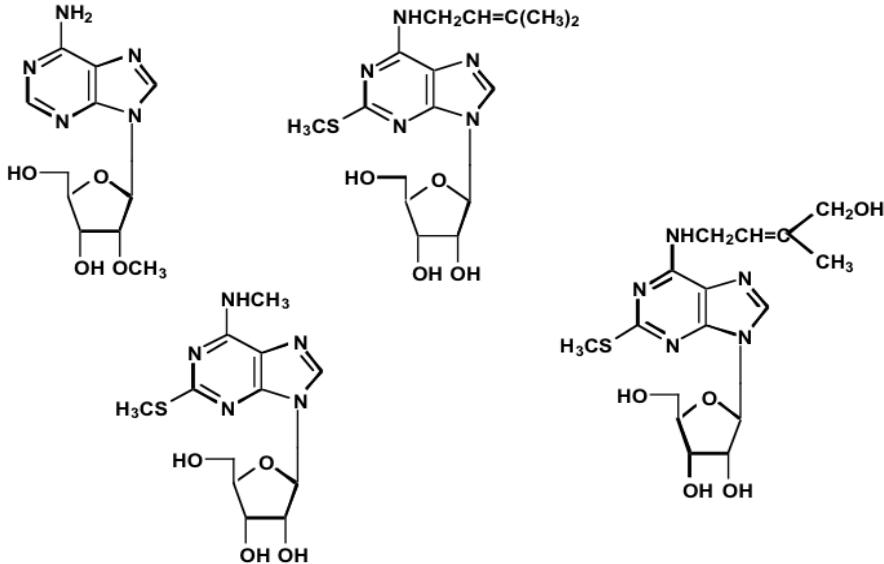


Nucleosides

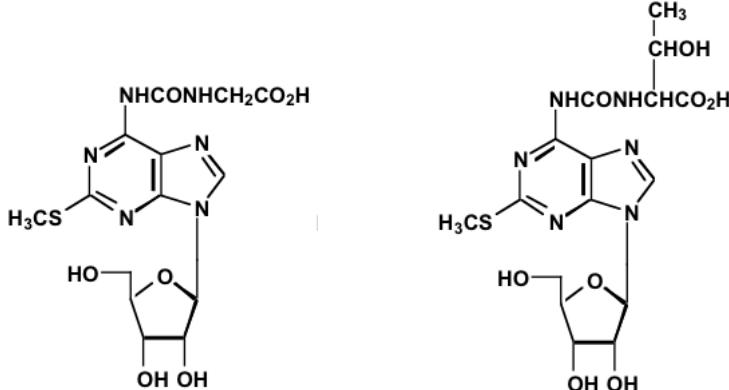
- Nucleosides are not just adenosine, cytosine, guanosine or uridine
 - “Different strokes for different folks”



More hydrophobic nucleosides



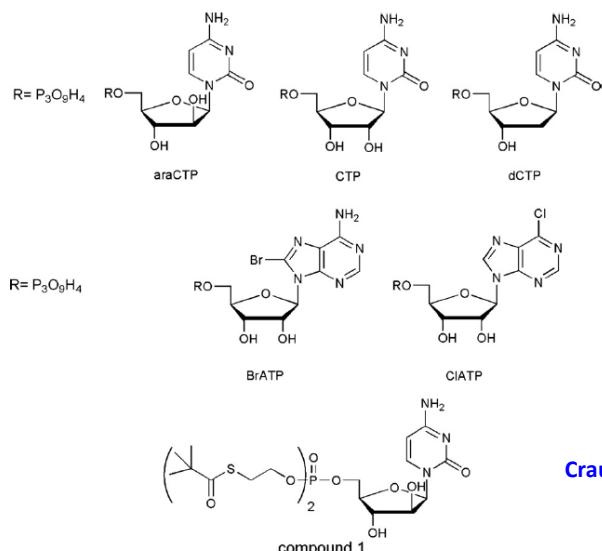
Hydrophilic nucleosides



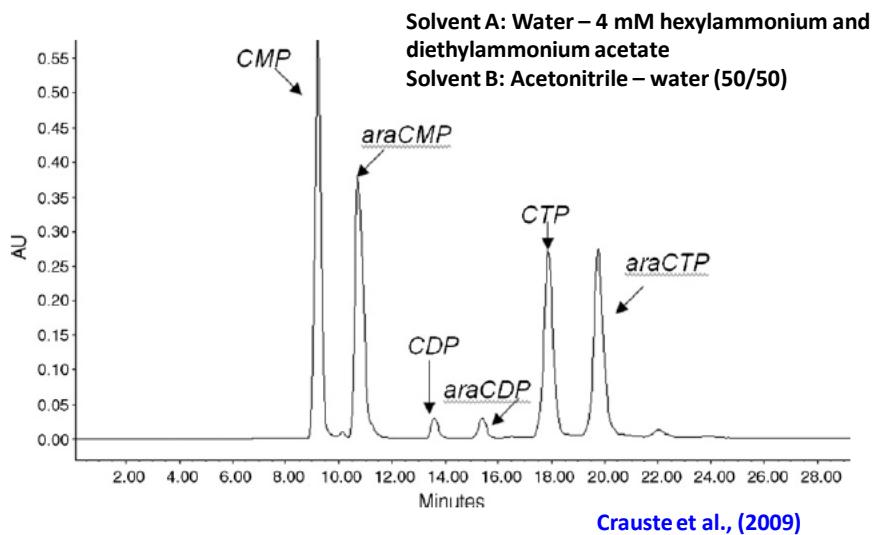
Nucleotides

- Important to convert them to ammonium, not Na^+ salts
 - DNA/RNA reprecipitated from 2 M ammonium acetate by alcohol at -20°C
- Nucleotides can be analyzed by LC-MS using a hydrophobic counter ion
 - E.g., tetrabutyl ammonium acetate (TBAA)
- Both NH_4^+ and TBAA salts dissociate in the ESI interface

CTP and ATP derivatives



LC-UV of NTPs

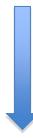


Sample prep for NTPs from cells

Washed cells extracted with 60% aqueous methanol and kept at -20°C overnight



Extract centrifuged at 13,200 xg for 3 minutes at RT



Supernatant diluted in 60% aqueous methanol and passed over a weak anion exchange column.

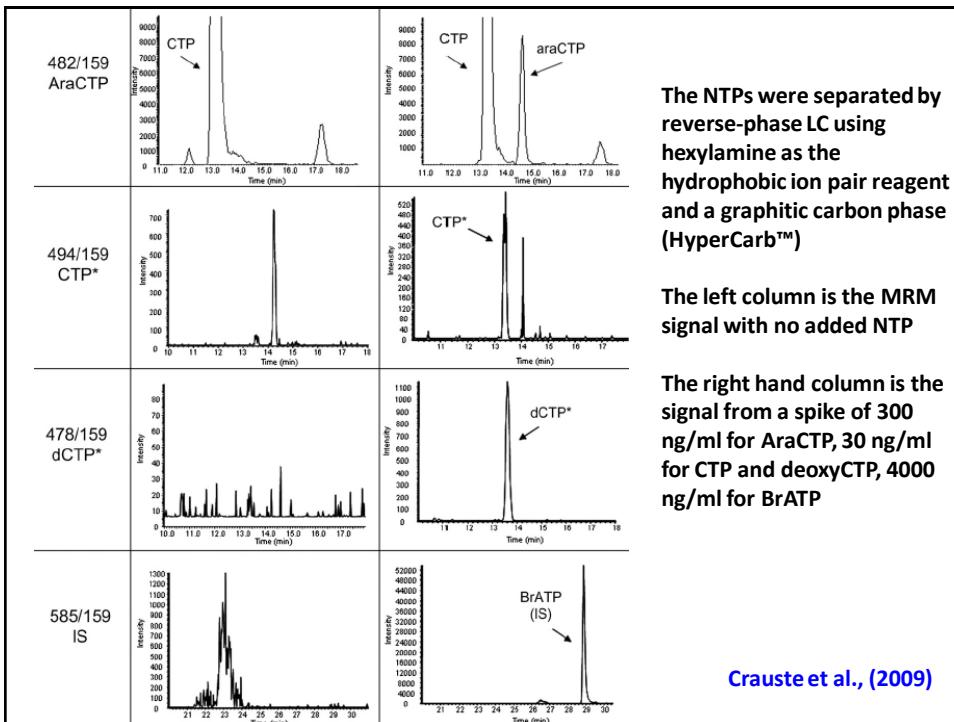


NTPs eluted with methanol-water-ammonia (24:5:1)

Evaporated to dryness and reconstituted in LC starting buffer.

Column washed with 50 mM ammonium acetate, pH 4.5

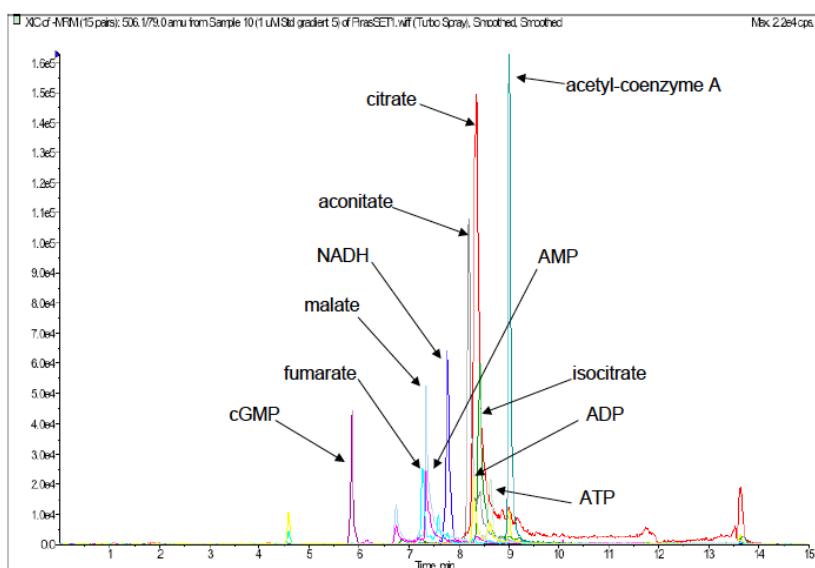
Crauste et al., (2009)



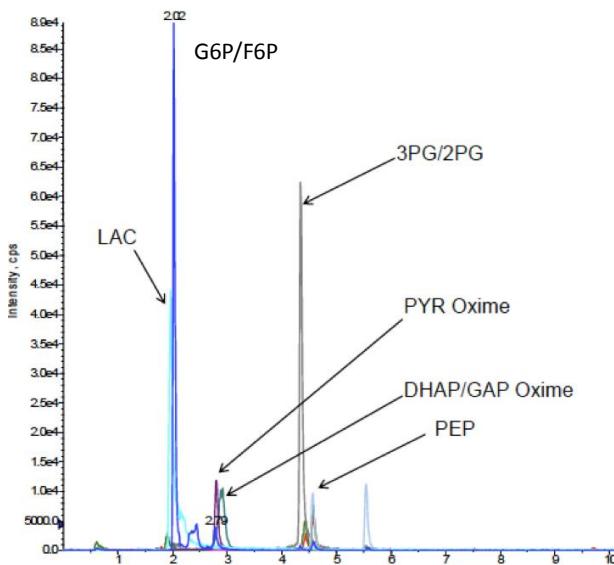
High energy intermediates

- Hexose and triose phosphates as well as Krebs cycle intermediates
 - These are very hydrophilic compounds
- Do not separate well on a reverse-phase column
 - Good separation in tributylammonium and better in tetrabutylammonium acetate
- “Naked” molecular ions

Ion-pair LC-MRM-MS of energy intermediates



Glycolytic intermediates by LC-MS



Suggested reading (ESI)

- Crauste C, Lefebvre I, Hovaneissian M, Puy JY, Roy B, Peyrottes S, Cohen S, Guittot J, Dumontet C, Perigaud C. *J Chromatogr B Analyt Technol Biomed Life Sci.* 2009;877:1417-1425.
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Suggested reading (MALDI)

- Tost J, Gut IG. *J Mass Spectrom* 2006; 41:981-995.
- Berkenkamp S, Kirpekar F, Hillenkamp F. *Science.* 1998;281:260-262.
- Mengel-Jørgensen J, Sanchez JJ, Børsting C, Kirpekar F, Morling N. *Anal Chem.* 2004;76:6039-6045.
- Keough T, Shaffer JD, Lacey MP, Riley TA, Marvin WB, Scurria MA, Hasselfield JA, Hesselberth EP. *Anal Chem.* 1996;68:3405-3412.